

(b) Action of Halides of phosphorus on alcohol:

(i) PX_3 (PCl_3 , PBr_3 , PI_3) $\rightarrow$



$\rightarrow$ This will happen thrice till all elements halogens from phosphorus and to replace 'OH' are removed

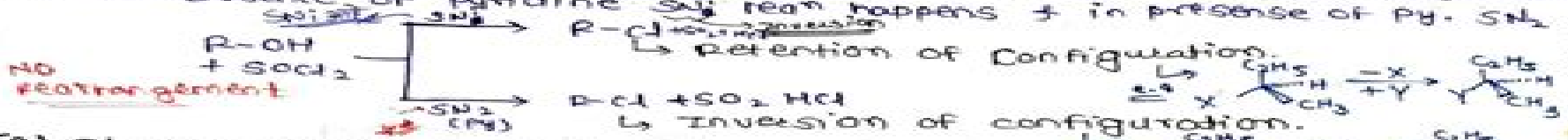
(ii) PCl_5 $\rightarrow$



(c) Action of SO_2Cl_2 on alcohol:



- $\rightarrow$ Only SO_2Cl_2 exist, $SOBr_2$ & SOI_2 don't.
- $\rightarrow$ This is the best method for preparation of alkyl chloride.
- $\rightarrow$ Also called as "Darzens reaction" $\because$ all products formed are gaseous in nature & escape out. Best method
- $\rightarrow$ In absence of pyridine SN_1 rean happens & in presence of Py , SN_2



(2) From alkenes by hydrohalogenation $\rightarrow$

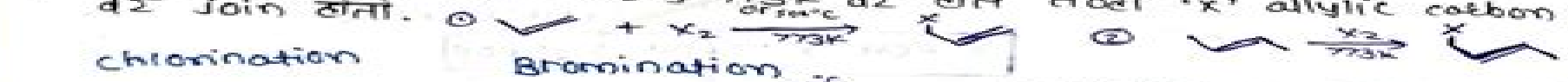


• Exceptions to Markovnikov Rule $\rightarrow$



- $\rightarrow$ "HBr + peroxide" Undergoes anti-Markovnikov (or) Kharasch-myo effect (or) peroxide effect. $\rightarrow$ [Only for HBr not for HCl, H_2I]
- $\rightarrow$ From this rule, Br should be attached to that carbon which is having more hydrogens. $\rightarrow$ Rearrangement possible.

$\rightarrow$ अगर अल्केन सी रीरन 'x' & 773K पर होते है तो अल्केन 'x' allylic carbon पर जोड़ जाता है।

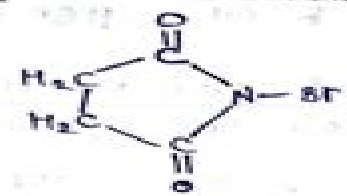


Chlorination

Bromination



• NBS [N-bromosuccinimide]



$\rightarrow$ chlorination is faster than bromination. $\rightarrow \therefore$ chlorine is more reactive than bromine

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