

Nuclear Chemistry Worksheet

Part A. Identify the following as alpha, beta, gamma, or neutron.

1. 1_0n _____
2. ${}^0_{-1}e$ _____
3. ${}^4_2\text{He}$ _____
4. ${}^0_0\gamma$ _____
5. Nuclear decay with no mass and no charge _____
6. An electron _____
7. Least penetrating nuclear decay _____
8. The most damaging nuclear decay to the human body _____
9. Nuclear decay that can be stopped by skin or paper _____
10. Nuclear decay that can be stopped by aluminum _____

Part B. Complete the following nuclear equations.

11. ${}^{40}_{19}\text{K} \rightarrow {}^0_{-1}e + \underline{\hspace{2cm}}$
12. ${}^{239}_{94}\text{Pu} \rightarrow {}^4_2\text{He} + \underline{\hspace{2cm}}$
13. ${}^{40}_{19}\text{K} \rightarrow {}^0_{-1}e + \underline{\hspace{2cm}}$
14. ${}^{238}_{92}\text{U} \rightarrow \underline{\hspace{2cm}} + {}^{234}_{90}\text{Th}$
15. ${}^6_3\text{Li} \rightarrow {}^4_2\text{He} + \underline{\hspace{2cm}}$
16. $\underline{\hspace{2cm}} \rightarrow {}^{142}_{56}\text{Ba} + {}^{91}_{36}\text{Kr} + 3 {}^1_0n$

Part C. Write the equations for the following processes.

17. The alpha decay of radon – 198
18. The beta decay of uranium – 237
19. Positron emission from silicon – 26
20. Sodium – 22 undergoes electron capture

Nuclear Chemistry Homework Sheet

SJ Ball



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