

COVALENT BONDING

Name _____

Covalent bonding occurs when two or more nonmetals share electrons, attempting to attain a stable octet of electrons at least part of the time. For example:

*H wants 2 electrons
all other atoms 8*



Note that hydrogen is content with 2, not 8, electrons.

electrons in the middle share

Show how covalent bonding occurs in each of the following pairs of atoms. Atoms may share one, two or three pairs of electrons.

wanted available share

1. $\text{H} + \text{H} (\text{H}_2)$ W - 4 A - 2 S - 2	<i>wanted 2 + 2</i> <i>available</i> $\text{H} \cdot \cdot \text{H}$ $\text{H} \cdot \quad \cdot \text{H}$	$\text{H} - \text{H}$ <i>single bond = share 2 e⁻</i>
2. $\text{F} + \text{F} (\text{F}_2)$ W - 8 + 8 = 16 A - 7 + 7 = 14 S - 2	$\cdot \ddot{\text{F}} \cdot \cdot \ddot{\text{F}} \cdot$ <i>2 e⁻</i>	$\cdot \ddot{\text{F}} - \ddot{\text{F}} \cdot$ <i>single bond 2 e⁻</i>
3. $\text{O} + \text{O} (\text{O}_2)$ W - 8 + 8 = 16 A - 6 + 6 = 12 S - 4	$\cdot \ddot{\text{O}} \cdot \cdot \ddot{\text{O}} \cdot$ <i>4 e⁻</i>	$\cdot \ddot{\text{O}} = \ddot{\text{O}} \cdot$ <i>4 double bond</i>
4. $\text{N} + \text{N} (\text{N}_2)$ W - 8 + 8 = 16 A - 5 + 5 = 10 S - 6	$\cdot \ddot{\text{N}} \cdot \cdot \ddot{\text{N}} \cdot$	$\cdot \ddot{\text{N}} \equiv \ddot{\text{N}} \cdot$
5. $\text{C} + \text{O} (\text{CO}_2)$ W - 8 + 8 + 8 = 24 A - 4 + 6 + 6 = 16 S - 8	$\cdot \ddot{\text{O}} \cdot \cdot \ddot{\text{C}} \cdot \cdot \ddot{\text{O}} \cdot$	$\cdot \ddot{\text{O}} = \text{C} = \ddot{\text{O}} \cdot$
6. $\text{H} + \text{O} (\text{H}_2\text{O})$ W - 2 + 2 + 8 = 12 A - 1 + 1 + 6 = 8 S - 4	$\text{H} \cdot \cdot \ddot{\text{O}} \cdot \cdot \text{H}$	$\text{H} - \ddot{\text{O}} - \text{H}$

Note Taking Guide Covalent Bonding Answer Key

**Patrick E. McMahon, Bohdan B.
Khomtchouk, Claes Wahlestedt**



Note Taking Guide Covalent Bonding Answer Key:

Reading and Writing in Science Maria C. Grant, Douglas Fisher, Diane Lapp, 2015-01-21 Engage your students in scientific thinking across disciplines Did you know that scientists spend more than half of their time reading and writing Students who are science literate can analyze present and defend data both orally and in writing The updated edition of this bestseller offers strategies to link the new science standards with literacy expectations and specific ideas you can put to work right away Features include A discussion of how to use science to develop essential 21st century skills Instructional routines that help students become better writers Useful strategies for using complex scientific texts in the classroom Tools to monitor student progress through formative assessment When students are curious they thrive Give your students the strong base they need to create and share scientific ideas that have an impact in the classroom and beyond This is a teacher friendly book that drew me in from the introduction to the end Through real life scenarios combined with useful methods for instruction it illustrates how science teachers can use language as a tool for teaching science Trina Allen Science Content Specialist Measurement Incorporated An eminently readable guide for the novice and experienced teacher The many practical ideas in this volume demonstrate that improving students skills in reading and writing can also improve their understanding and ability in science Cary Sneider Associate Research Professor Portland State University Portland OR

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Basic Bioscience Laboratory Techniques Philip L.R. Bonner, Alan J. Hargreaves, 2011-08-24 This unique practical pocket sized guide and reference provides every first year bioscience student with all they need to know to prepare reagents correctly and perform fundamental laboratory techniques It also helps them to analyse their data and present their findings in addition to directing the reader via a comprehensive list of references to relevant further reading All of the core bioscience laboratory techniques are covered including basic calculations and the preparation of solutions aseptic techniques microscopy techniques cell fractionation spectrophotometry chromatography of small and large molecules electrophoresis of proteins and nucleic acids and data analysis In addition the book includes clear relevant diagrams and worked examples of calculations In short this is a must have for all first year bioscience students struggling to get to grips with this vitally important element of their course

Survival Guide to Organic Chemistry Patrick E. McMahon, Bohdan B. Khomtchouk, Claes Wahlestedt, 2016-12-19 The Survival Guide to Organic Chemistry Bridging the Gap from General Chemistry enables organic chemistry students to bridge the gap between general chemistry and organic chemistry It makes sense of the myriad of in depth concepts of organic chemistry without overwhelming them in the necessary detail often given in a complete organic chemistry text Here the topics covered span the entire standard organic chemistry curriculum The authors describe subjects which require further explanation offer alternate viewpoints for understanding and provide hands on practical problems and solutions to help master the material This text ultimately allows students to apply key ideas from their general chemistry curriculum to key concepts in organic chemistry

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Reviews key general chemistry concepts and techniques adapted for application to important organic principles Provides practical guidance to help students make the notoriously well known and arduous transition from general chemistry to organic chemistry Explains organic concepts and reaction mechanisms generally expanding the focus on how to understand each step from a more intuitive viewpoint Covers concepts that need further explanation as well as those that summarize and emphasize key ideas or skills necessary in this field An added bonus is help with organizing principles to make sense of a wide range of similar reactions and mechanisms Implements a user friendly process to achieve the end result of problem solving Covers organic chemistry I and II concepts at the level and depth of a standard ACS organic chemistry curriculum features practice problems and solutions to help master the material including an extensive and comprehensive bank of practice exams with solutions

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