

A crystalline solid is formed when 2 or more atoms exchange valence electrons to increase stability.

Ionic bond is formed when valence electrons are transferred from one atom to another.

cations = atoms lose electrons to become [+] charged cations.

anions gain electrons to become [-] charged.

In ionic compounds that ions are arranged in a crystal lattice. Strong forces hold the ions together.

Properties of ionic compounds:

- = high melting and boiling points
- = hard & not easily crushed
- = conduct electricity when melted or dissolved because the ions are free to move

covalent bond = electrons are shared. Forming molecules

Covalent compounds have weak forces holding the molecules together.

Properties of covalent compounds:

- lower melting and boiling points
- Many covalent compounds are volatile liquids or gases.
- soft, easier to crush
- are not conductors of electricity
- electronegativity = property that tells how strong an

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George William MacArthur Reynolds



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The Odyssey Close Reading and Text-Dependent Questions Jennifer Kroll, 2014-10-01 Students analyze The Odyssey using key skills for college and career readiness Close reading of the text is required to answer text dependent questions Included are student pages with the text dependent questions as well as suggested answers Agricultural Library Notes ,1942

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