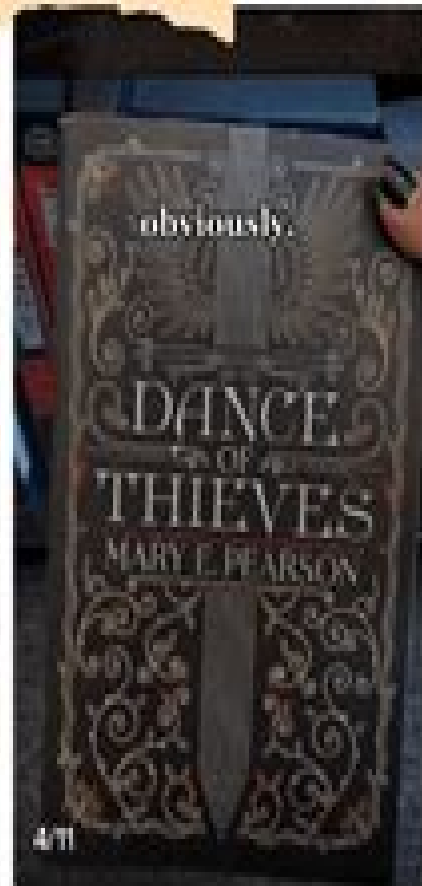


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web procedure 1 find the mass of a clean dry 100 ml beaker and record 2 measure out 8 00 grams of copper ii sulfate onto a piece of filter paper taking the filter paper s mass into account add these to the beaker 3 measure 50 0 ml of water in the cylinder and add it to the beaker 4

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the reaction between 7.0002g away copper ii sulfate and 2.0101g iron metal which produced 2.3037g of copper product with a percent yield of 100.72 introduction

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