

ACS ORGANIC CHEMISTRY FINAL EXAM 2

- How can the free-radical addition reaction of

1. HBr + alkene + peroxides

2. the polymerization of alkenes or dienes be *initiated*?: heat, peroxides, or UV light

- What is the trend in stability for organic radicals/carbocations?: allylic/ben-zylic > 3 alkyl > 2 alkyl > 1 alkyl > methyl > vinylic/aryl

*Allylic and benzylic radicals are especially easy to generate because of the stabilization caused by resonance.

* when formed in a ring, the C+ will form preferentially to the C that can be stabilized by resonance.

- What conditions will usually indicate a radical reaction?: AIBN (a radical initiator), peroxides, UV light, or high temps (300-500 C)

• Groups that reduce electron density of a ring by resonance or induction will: cause the ring to be less reactive than benzene

• Groups that increase the electron density of the ring by resonance or induction will: be more reactive than benzene

• Electron-donating groups cause the electron density to increase at the: ortho (next to) and para (opposite) positions. Causes electrophiles to prefer to attack at those positions

• Electron withdrawing groups cause electron density to be reduced at the: ortho and para position more than at the meta position. Causes electrophiles to prefer to attack at the meta position.

- What has the greatest influence on electron density of an aromatic ring?-

: Resonance

• What increases electron density in an aromatic ring?: Groups that have an atom with non bonded electrons

• What reduces electron density?: groups that have multiple bonds to an atom more electronegative than carbon

- What are the two distinct pathways for nucleophilic aromatic substitution?-

: Addition-elimination and elimination-addition.

- What is the addition-elimination mechanism?: 1. aromatic ring accepts a pair of electrons

Organic Chemistry Final Exam Study Guide

Jill Meryl Levy



Organic Chemistry Final Exam Study Guide:

Study Guide and Solutions Manual for Fundamentals of Organic Chemistry Susan McMurry, 1994 Following a brief review of structure and bonding organic molecules and functional groups are presented as early as possible The text is organized primarily by functional group beginning with simple alkanes and moving toward more complex compounds Emphasis is placed on the fundamental mechanistic similarities of organic reactions McMurry's thorough revision continues to present the solid content necessary for this course without sacrifice of important subjects and pedagogical tools Text and reaction summaries full problem sets and outstanding artwork are just some of the features in the Third Edition usually found in a full year book McMurry's clear well written explanations remain a highlight of the book

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[Hazmat Chemistry Study Guide \(Second Edition\)](#) Jill Meryl Levy, 2011 *Organic Chemistry Education Research into Practice* Jay

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University of Michigan Official Publication University of Michigan,1974 Each number is the catalogue of a specific school or college of the University Organic Chemistry David R. Klein,2017-08-14 In Organic Chemistry 3rd Edition Dr David Klein builds on the phenomenal success of the first two editions which presented his unique skills based approach to learning organic chemistry Dr Klein s skills based approach includes all of the concepts typically covered in an organic chemistry textbook and places special emphasis on skills development to support these concepts This emphasis on skills development in unique SkillBuilder examples provides extensive opportunities for two semester Organic Chemistry students to develop proficiency in the key skills necessary to succeed in organic chemistry *Creating Wicked Students* Paul Hanstedt,2023-07-03 In *Creating Wicked Students* Paul Hanstedt argues that courses can and should be designed to present students with what are known as wicked problems because the skills of dealing with such knotty problems are what will best prepare them for life after college As the author puts it this book begins with the assumption that what we all want for our students is that they be capable of changing the world When a student leaves college we want them to enter the world not as drones participating mindlessly in activities to which they ve been appointed but as thinking deliberative beings who add something to society There s a lot of talk in education these days about wicked problems problems that defy traditional expectations or knowledge problems that evolve over time Zika ISIS political discourse in the era of social media To prepare students for such wicked problems they need to have wicked competencies the ability to respond easily and on the fly to complex challenges Unfortunately a traditional education that focuses on content and skills often fails to achieve this sense of wickedness Students memorize for the test prepare for the paper practice the various algorithms over and over again but when the parameters or dynamics of the test or the paper or the equation change students are often at a loss for how to adjust This is a course design book centered on the idea that the goal in the college classroom in all classrooms all the time is to develop students who are not just loaded with content but capable of using that content in thoughtful deliberate ways to make the world a better place Achieving this goal requires a top to bottom reconsideration of courses including student learning goals text selection and course structure day to day pedagogies and assignment and project design *Creating Wicked Students* takes readers through each step of the process providing multiple examples at each stage while always encouraging instructors to consider concepts and exercises in light of their own courses and students

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