

The diagram illustrates the Calvin cycle, a series of biochemical reactions that convert carbon dioxide into glucose. The cycle is divided into three main stages:

- Carbon Fixation (Stage 1):** Carbon dioxide (CO_2) enters the cycle and combines with ribulose biphosphate (RuBP) to form oxaloacetate. This reaction is catalyzed by the enzyme RuBisCo.
- Reduction (Stage 2):** Oxaloacetate is converted into malate, which is then converted into pyruvate. This stage involves the use of ATP and NADPH, which are converted back to ADP and NADP⁺.
- Carbohydrate Formation (Stage 3):** Pyruvate is converted into glucose. This stage also involves the use of ATP and NADPH, which are converted back to ADP and NADP⁺.

The diagram also shows the regeneration of RuBP from pyruvate, completing the cycle. The final products are glucose and oxygen (O_2).

Linear polymer

Polymerization

Branched polymer

Polymerization

Linear polymers are formed by the addition of monomers to a growing chain. The monomers are added to the chain in a specific order, which determines the structure of the polymer.

Branched polymers are formed by the addition of monomers to a growing chain. The monomers are added to the chain in a specific order, which determines the structure of the polymer.

Linear polymers are formed by the addition of monomers to a growing chain. The monomers are added to the chain in a specific order, which determines the structure of the polymer.

Branched polymers are formed by the addition of monomers to a growing chain. The monomers are added to the chain in a specific order, which determines the structure of the polymer.

- Constitutional Isomers** = completely different molecules
- Conformational Isomers** = Same molecules that make different 3D shapes, same all free rotation around single bonds
- Enantiomers** = mirror image molecules, cannot be superimposed, same, having physical & chemical properties in achiral environment
- Diastereomers** = mirror image molecules, cannot be superimposed, have different physical & chemical properties
- Meso Compounds** = non-chiral due to the plane of symmetry

[illegible]

Metabolism:

1. **Pyruvate to Acetyl-CoA:** Pyruvate is converted to Acetyl-CoA in the mitochondria. This process involves the loss of CO₂ and the addition of CoA. The reaction is catalyzed by the Pyruvate Dehydrogenase Complex (PDH complex).

2. **Acetyl-CoA Entry:** Acetyl-CoA enters the Citric Acid Cycle by combining with Oxaloacetate to form Citrate. This reaction is catalyzed by Citrate Synthase.

3. **Citrate Conversion:** Citrate is converted to Isocitrate by the enzyme Isocitrate Dehydrogenase. This step involves the loss of CO₂ and the reduction of NAD⁺ to NADH.

4. **Isocitrate to α-Ketoglutarate:** Isocitrate is converted to α-Ketoglutarate by the enzyme α-Ketoglutarate Dehydrogenase. This step involves the loss of CO₂ and the reduction of NAD⁺ to NADH.

5. **α-Ketoglutarate to Succinyl-CoA:** α-Ketoglutarate is converted to Succinyl-CoA by the enzyme Succinyl-CoA Synthetase. This step involves the loss of CO₂ and the production of GTP (or ATP).

6. **Succinyl-CoA to Succinate:** Succinyl-CoA is converted to Succinate by the enzyme Succinate Dehydrogenase. This step involves the reduction of FAD to FADH₂.

7. **Succinate to Malate:** Succinate is converted to Malate by the enzyme Succinate Dehydrogenase. This step involves the reduction of FAD to FADH₂.

8. **Malate to Oxaloacetate:** Malate is converted to Oxaloacetate by the enzyme Malate Dehydrogenase. This step involves the oxidation of NADH to NAD⁺.

9. **Regeneration of Oxaloacetate:** Oxaloacetate is regenerated from Malate, completing the cycle.

Key Intermediates and Cofactors:

- Pyruvate
- Acetyl-CoA
- Citrate
- Isocitrate
- α-Ketoglutarate
- Succinyl-CoA
- Succinate
- Malate
- Oxaloacetate
- NAD⁺ / NADH
- FAD / FADH₂
- GTP / ATP

Glucose (6C) → Glucose-6-phosphate (6C) → Fructose-1,6-bisphosphate (6C) → Glyceraldehyde-3-phosphate (3C) → Pyruvate (3C)

Enzymes: Hexokinase, Phosphofructokinase, Pyruvate Kinase

Pathway: Aerobic, Anaerobic

The diagram illustrates the electrophilic aromatic substitution mechanism. It starts with a benzene ring reacting with an electrophile (E⁺). The reaction is labeled "Electrophilic Substitution" and "Electrophile". The mechanism proceeds through a sigma complex intermediate, which is shown in brackets. The sigma complex is a cyclohexadienyl cation with a positive charge delocalized over five carbons. The final step shows the loss of a proton to restore aromaticity, yielding the substituted benzene product.

Polystyrene Copolymerizing groups (-CH₂-CH-)

Common: found in all developing groups in that they
 1. grow to maturity with no change in feeding or nesting habits

1. **Identify the main idea of the passage.**

100

[illegible][illegible]

Reaction scheme showing the conversion of a cyclic acetal to a cyclic ketal. The starting material is a cyclic acetal (a central carbon atom bonded to two methoxy groups and two ethoxy groups). It reacts with a ketone (acetone) in the presence of an acid catalyst (H+) to form a cyclic ketal. The product is a cyclic ketal where the central carbon atom is bonded to two methoxy groups and two ethoxy groups, and the ketone oxygen is now part of the ring structure.

 chloroplast (in plant)

 chloroplast (in animal)

The diagram illustrates the process of group polarization. It shows a group of individuals (represented by circles) interacting and influencing each other, leading to a more extreme position (represented by a circle with a star) than the initial position (represented by a circle with a dot). The diagram is labeled "group polarization" and "group of individuals".

Organic Chemistry Semester 1 Mechanism Study Guide

Michael B. Smith, Jerry March



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