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best completes each statement or best answers each question 1 which of the following is an oxyacid a hcl b h₂o c h₂s d h₂so₄
2 which acid is produced in the stomach a hno₃ b ch₃cooh c

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web apr 20 2016 a stronger acid c weaker acid b stronger base d weaker base 6 proton transfer reactions favor production
of the a stronger acid and stronger base b weaker acid and weaker base c stronger acid and weaker base d weaker acid and
stronger base 7 aqueous solutions of most bases contain a hydroxide ions and

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web correct answer d ccl 3 cooh explanation strong acid means weak conjugate base ccl 3 coo chcl 2 coo ch₂clcoo ch 3 coo

therefore the strongest acid is $\text{C}_6\text{H}_5\text{COOH}$ q2 lime water is a Ca(OH)_2 c CaCO_3 d CaCl_2 correct answer b Ca(OH)_2
explanation

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 $\text{aq C}_3\text{H}_7\text{NO}_2 + \text{aq H}_3\text{O}^+ \rightleftharpoons \text{aq C}_3\text{H}_8\text{NO}_2^+ + \text{aq H}_2\text{O}$ l $\text{CH}_3\text{CO}_2\text{H} + \text{aq NH}_3 \rightleftharpoons \text{aq CH}_3\text{CO}_2^- + \text{aq NH}_4^+$ aq $\text{SbF}_5 + \text{aq 2HF} \rightleftharpoons \text{aq SbF}_6^- + \text{aq 2H}^+$ identify the conjugate acid base

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web jul 22 2023 correct answer c KH_2PO_4 explanation KH_2PO_4 is an acid salt because it is derived from a weak acid phosphoric acid and a strong base potassium hydroxide it contains the hydrogen ion from the acid making it acidic but also has the ability to form basic ions due to the presence of the potassium ion

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web the ionization of weak acids and weak bases many acids and bases are weak that is they do not ionize fully in aqueous solution a solution of a weak acid in water is a mixture of the nonionized acid hydronium ion and the conjugate base of the acid with the nonionized acid present in the greatest concentration

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web oct 10 2022 correct answer d salt water explanation neutralization is a chemical reaction between an acid and a base resulting in the formation of a salt and water the product of neutralization is therefore salt water as the acid and base

combine to form a salt compound and water molecules rate this question 4