Oxidation Numbers

Summary: An atom connected to a more eN atom = +1 An atom connected to a less eN atom=-1 An atom connected the same atom = 0 *Number of bonds must be taken into account * Charge is also incorporated into oxidation number

Oxidation number 1 = Oxidized = reducing agent Oxidation number + = reduced = oxidizing agent 00 - - - 2

Examples: 0 = C = 0

* Charge is added

Oxidation Reactions

Summary: 10 alcohol 3 aldehyde Har > carboxylic acid

lo alconol col aldenyde must be dry 2° alcohol Co> ketone can be either wet ordry

3° alconol 603 no further oxidation --- OH

Overall mechanism:

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Jiben Roy

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