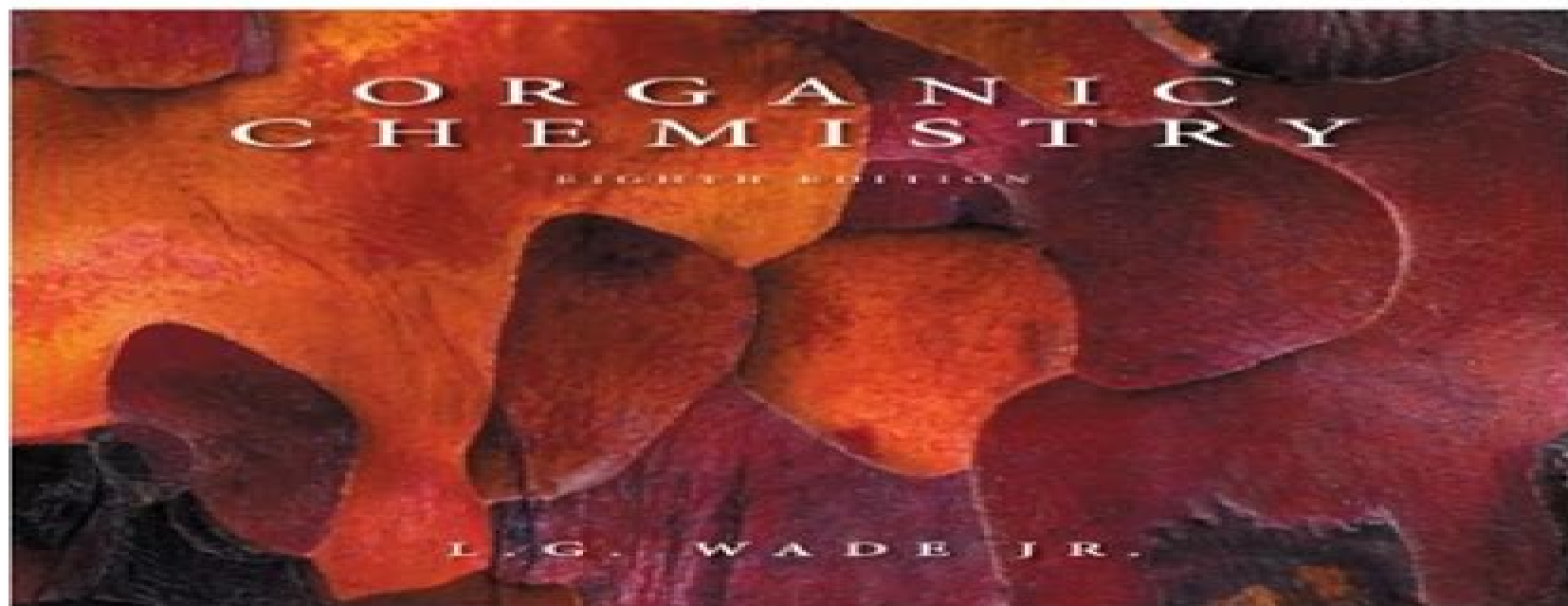
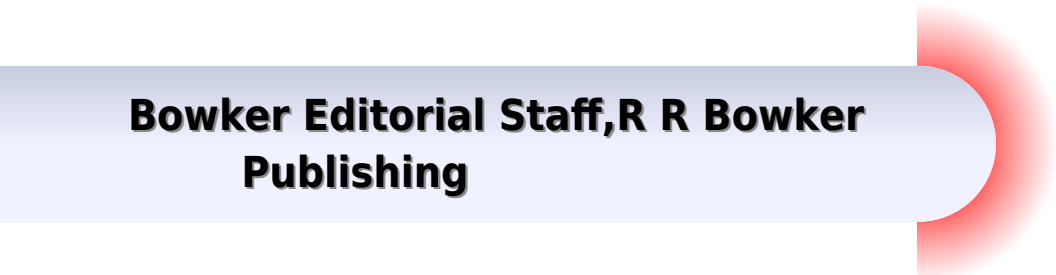


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web we can write a balanced nuclear equation for this reaction as let s do a practice example identify the product x by writing a balanced nuclear equation ${}^{137}_{55}\text{Cs} \rightarrow {}^{137}_{56}\text{Ba} + x$ because the mass number is the same on both sides of the equation x must have a

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212 beryllium 8 and a positron are produced by the decay of an unstable nucleus neptunium 239 forms from the reaction of uranium 238 with a neutron and then spontaneously converts into plutonium 239 strontium 90 decays into

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