

A guide to formulae in organic chemistry

Molecular formula

The molecular formula of an organic compound simply shows the number of each type of atom present. It tells you nothing about the bonding within the compound.



Molecular formula of butanoic acid

Empirical formula

The empirical formula of an organic compound gives the simplest possible whole number ratio of the different types of atom within the compound.



Empirical formula of butanoic acid

Condensed formula

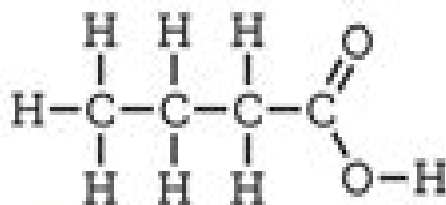
In condensed formulas, each carbon atom is listed separately, with atoms attached to it following. In cyclic parts of molecules, like benzene, carbons are grouped.



Condensed formula of butanoic acid

Displayed formula

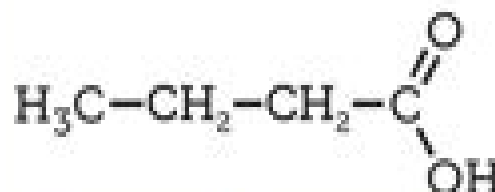
A displayed formula shows all of the atoms and all of the bonds present in an organic compound. The bonds are represented as lines.



Displayed formula of butanoic acid

Structural formula

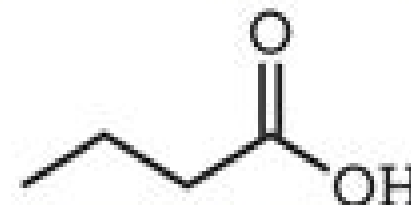
Similar to displayed formula, not all bonds are shown, although all atoms are still indicated using subscript numbers. Carbon hydrogen bonds are often simplified.



Structural formula of butanoic acid

Skeletal formula

In skeletal formulas, most hydrogen atoms are omitted. Line ends or vertices represent carbons. Functional groups and atoms other than carbon or hydrogen are still shown.



Skeletal formula of butanoic acid

Organic Structural Formula Practice

RD Boyd



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Processing and Information Systems Birger Andersson, Maria Bergholtz, P. Johannesson, 2003-07-01 The workshop on Applications of Natural Language to Information Systems NLDB

has since 1995 provided a forum for academic and industrial researchers and practitioners to discuss the application of natural language to both the development and use of software applications

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Boyle, Andy Morse, 2014-10-13 Science for Geography and Environment introduces students and academics who need to brush up their knowledge to scientific principles in a lively and accessible way allowing them to proceed through the text at their own pace The book is structured thematically with a logical development of key topics all linked by a comprehensive cross referencing system Concepts and principles will be grounded in everyday experience and exemplified by reference to geographical environmental processes The authors are also testing each stage of the text on their own students thereby ensuring that student needs are given top priority in the book's development Lively and relevant introduction to those scientific principles necessary to understand key processes occurring within the natural environment **Practice Makes**

Perfect: Organic Chemistry Marian DeWane, Thomas J. Greenbowe, 2016-12-02 The equation for success in chemistry is practice practice practice Increase your confidence in your science skills with Practice Makes Perfect Organic Chemistry Inside are numerous lessons to help you better understand the subject These lessons are accompanied by exercises to practice what you've learned along with a complete answer key to check your work Throughout this book you will learn the critical terms to help you understand organic chemistry and you will expand your knowledge of the subject through dozens of sample problems and their solutions With the lessons in this book you will find it easier than ever to grasp concepts in organic chemistry And with a variety of exercises for practice you will become comfortable using your growing science skills in your classwork and on exams You'll be on your way to mastering these topics and more Properties of organic compounds Molecular geometry and isomers Nomenclature and structure of alcohols Reactions of carboxylic acids Acid base reactions Formation of esters Offering a winning method for comprehending essentials right away this book will become your indispensable resource for understanding the fundamentals of this science

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