

Oxidation Numbers

Summary: An atom connected to a **more** eN atom = +1

An atom connected to a **less** eN atom = -1

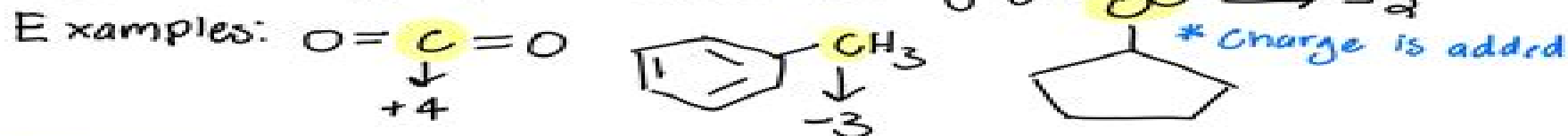
An atom connected the **same** atom = 0

* Number of bonds must be taken into account

* Charge is also incorporated into oxidation number

Oxidation number $\uparrow$ = Oxidized = reducing agent

Oxidation number $\downarrow$ = reduced = oxidizing agent



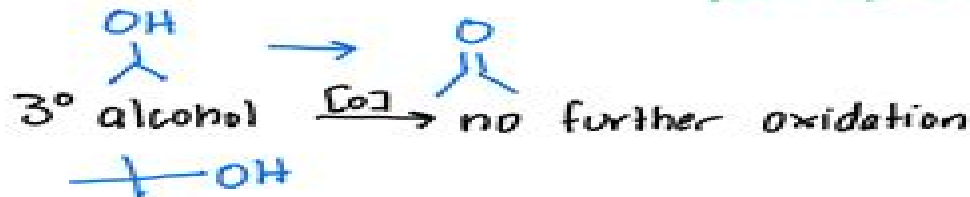
Oxidation Reactions

Summary: 1° alcohol $\xrightarrow{[\text{O}]}$ aldehyde $\xrightarrow[\text{H}_2\text{O}]{\text{must be wet } [\text{O}]}$ carboxylic acid

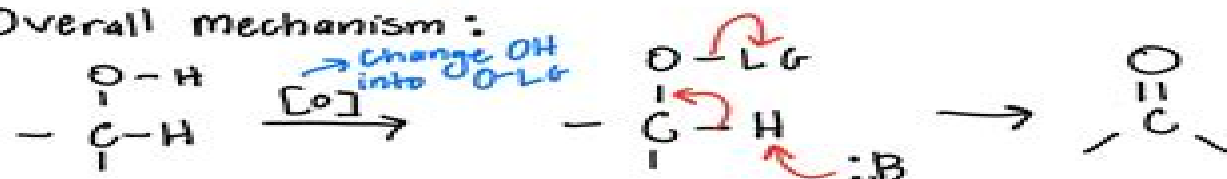


1° alcohol $\xrightarrow{[\text{O}]}$ aldehyde must be dry

2° alcohol $\xrightarrow{[\text{O}]}$ ketone can be either wet or dry



Overall mechanism:



Organic Chem Amine Study Guide

Geoff Neuss



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A Self-study Guide to the Principles of Organic Chemistry Jiben Roy, 2013 A Self Study Guide to the Principles of Organic Chemistry Key Concepts Reaction Mechanisms and Practice Questions for the Beginner will help students new to organic chemistry grasp the key concepts of the subject quickly and easily as well as build a strong foundation for future study Starting with the definition of atom the author explains molecules electronic configuration bonding hydrocarbons polar reaction mechanisms stereochemistry reaction varieties organic spectroscopy aromaticity and aromatic reactions biomolecules organic polymers and a synthetic approach to organic compounds The over one hundred diagrams and charts contained in this volume will help students visualize the structures and bonds as they read the text and make the logic of organic chemistry clear and easily understood Each chapter ends with a list of frequently asked questions and answers followed by additional practice problems Answers are included in the Appendix

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