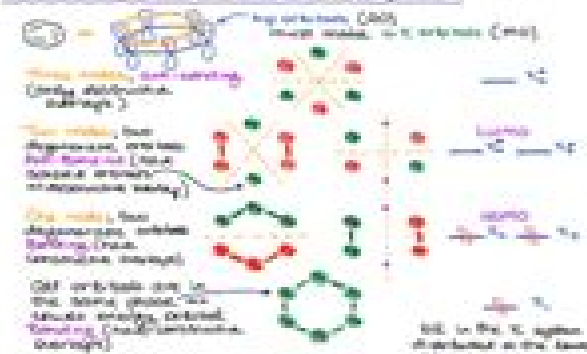
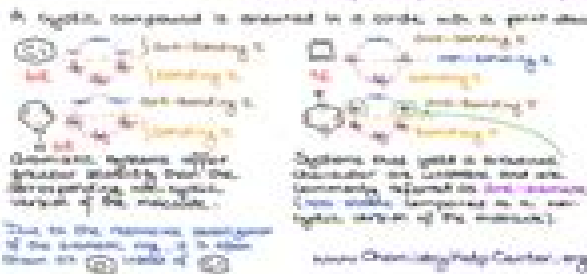


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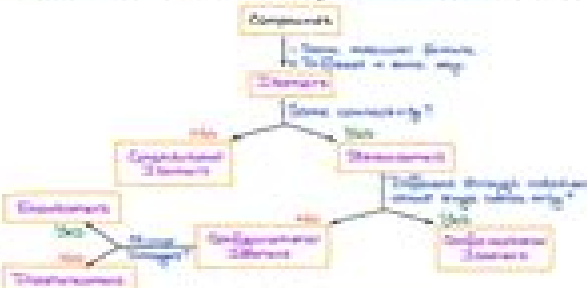


**Free Radical Reactions:** - free radicals are produced by homolysis of a bond, often initiated by heat or light.



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- Isomerism of Stereochemical Relationships:**
  - Constitutional Isomers:** - completely different molecules.
  - Conformational Isomers:** - same molecule that only differ in 3D shape due to free rotation around single bonds.
  - Enantiomers:** - non-superimposable mirror images that share physical & chemical properties in achiral environment.
  - Diastereomers:** - non-superimposable, non-mirror images that have different physical & chemical properties.
  - Meso Compounds:** - non-chiral due to the plane of symmetry, however, have chiral centers.



## Stereochemical Relationships of Enantiomers

Enantiomers are mirror images that are non-superimposable. They have identical physical properties in achiral environment, but differ in optical activity. They rotate plane polarized light in opposite directions.

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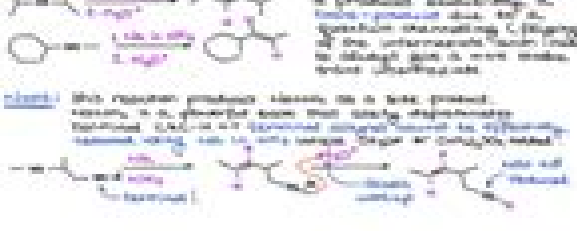
## Reaction of Aldehydes with Nucleophiles in Liquid Phase



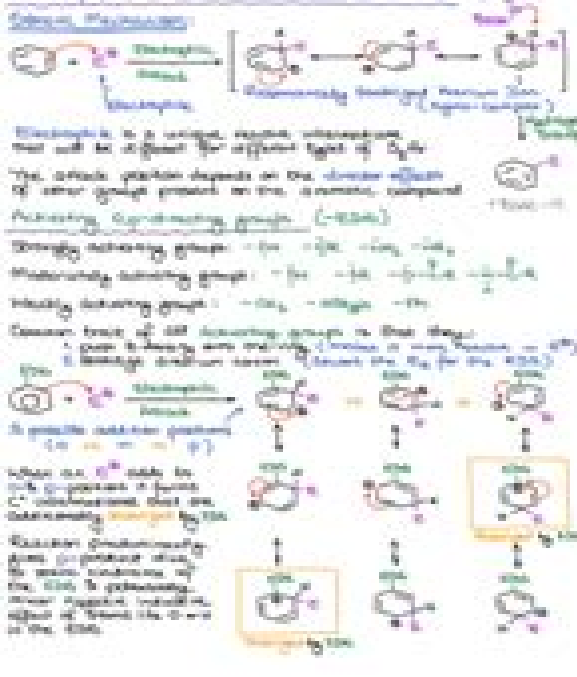
**Reaction of Aldehydes with Cyanide:**

1. **Nucleophilic addition:** The cyanide ion ( $CN^-$ ) attacks the carbonyl carbon.

2. **Protonation:** The intermediate is protonated to form the final cyanohydrin product.

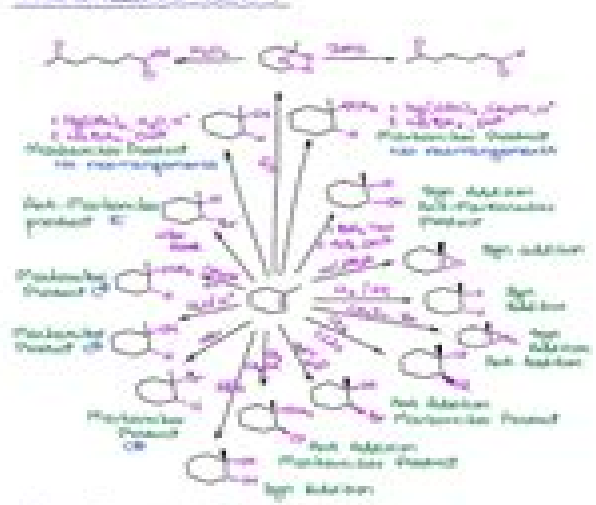


## Electrophilic Aromatic Substitution Reactions (EAS)



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## Alkene Reaction Mechanism



## Alkene Functional Group Conversion



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# Organic Chemistry Guide

**Paula Yurkanis Bruice**



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