

ORGANIC CHEMISTRY

REACTIONS

Reaction of alcohols with sodium

Alcohols react with sodium to form an alkoxide and hydrogen gas.

$$2\text{C}_2\text{H}_5\text{OH} + 2\text{Na} \rightarrow 2\text{C}_2\text{H}_5\text{O}^- \text{Na}^+ + \text{H}_2$$

The reaction is exothermic and the sodium metal is oxidized to sodium ions.

Reaction of alcohols with phosphorus pentachloride

Alcohols react with phosphorus pentachloride to form an alkyl chloride and phosphorus oxychloride.

$$\text{C}_2\text{H}_5\text{OH} + \text{PCl}_5 \rightarrow \text{C}_2\text{H}_5\text{Cl} + \text{POCl}_3$$

The reaction is exothermic and the phosphorus pentachloride is reduced to phosphorus oxychloride.

Reaction of alcohols with phosphorus pentachloride

Alcohols react with phosphorus pentachloride to form an alkyl chloride and phosphorus oxychloride.

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**Humira Sonah, Vinod Goyal, S. M.
Shivaraj, Rupesh K. Deshmukh**



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Table of Contents Organic Chemistry Reactions Quickstudy Academic

1. Understanding the eBook Organic Chemistry Reactions Quickstudy Academic
 - The Rise of Digital Reading Organic Chemistry Reactions Quickstudy Academic
 - Advantages of eBooks Over Traditional Books
2. Identifying Organic Chemistry Reactions Quickstudy Academic
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Organic Chemistry Reactions Quickstudy Academic
 - User-Friendly Interface

4. Exploring eBook Recommendations from Organic Chemistry Reactions Quickstudy Academic
 - Personalized Recommendations
 - Organic Chemistry Reactions Quickstudy Academic User Reviews and Ratings
 - Organic Chemistry Reactions Quickstudy Academic and Bestseller Lists
5. Accessing Organic Chemistry Reactions Quickstudy Academic Free and Paid eBooks
 - Organic Chemistry Reactions Quickstudy Academic Public Domain eBooks
 - Organic Chemistry Reactions Quickstudy Academic eBook Subscription Services
 - Organic Chemistry Reactions Quickstudy Academic Budget-Friendly Options
6. Navigating Organic Chemistry Reactions Quickstudy Academic eBook Formats
 - ePub, PDF, MOBI, and More
 - Organic Chemistry Reactions Quickstudy Academic Compatibility with Devices
 - Organic Chemistry Reactions Quickstudy Academic Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Organic Chemistry Reactions Quickstudy Academic
 - Highlighting and Note-Taking Organic Chemistry Reactions Quickstudy Academic
 - Interactive Elements Organic Chemistry Reactions Quickstudy Academic
8. Staying Engaged with Organic Chemistry Reactions Quickstudy Academic
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Organic Chemistry Reactions Quickstudy Academic
9. Balancing eBooks and Physical Books Organic Chemistry Reactions Quickstudy Academic
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Organic Chemistry Reactions Quickstudy Academic
10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Organic Chemistry Reactions Quickstudy Academic
 - Setting Reading Goals Organic Chemistry Reactions Quickstudy Academic
 - Carving Out Dedicated Reading Time

12. Sourcing Reliable Information of Organic Chemistry Reactions Quickstudy Academic
 - Fact-Checking eBook Content of Organic Chemistry Reactions Quickstudy Academic
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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