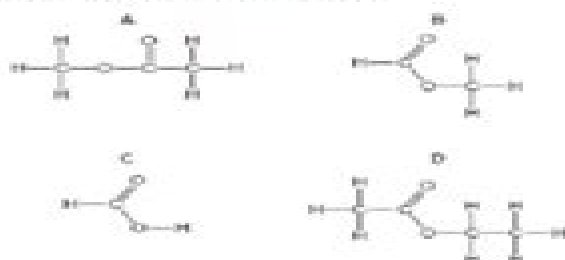


37 Which compound will react with ethanol to form an ester?



38 Which compound is manufactured by reacting ethene with steam in the presence of a catalyst?

- A** C_2H_6 **B** C_2H_5OH **C** C_2H_4 **D** C_2H_5OH

37 Which compound is obtained by the oxidation of ethanol, C_2H_5OH ?

- A** HCO_2CH_3
B $C_2H_5CO_2H$
C CH_3OH
D CH_3CO_2H

Ethanol can be manufactured from either ethene or glucose.

(i) The fermentation of glucose can be represented by the following equation.



Calculate the maximum mass of ethanol that could be made from 36 tonnes of glucose. [7]

(ii) Explain why ethanol made from ethene is a non-renewable fuel but that made from glucose is a renewable fuel. [2]

7. The diagram shows a reaction scheme.



What is the final compound, Z?

- A** a carboxylic acid
B an alcohol

12. Which conditions are necessary to ferment sugar into ethanol?

	product	temperature/ °C
A	absent	30
B	absent	70
C	present	30
D	present	70

8.

How does the number of carbon, hydrogen and oxygen atoms in an ester differ from the total number of carbon, hydrogen and oxygen atoms in the alcohol and carboxylic acid from which the compound was derived?

	Carbon atoms	Hydrogen atoms	Oxygen atoms
A	less	less	less
B	less	same	less
C	same	less	less
D	same	same	same

9.

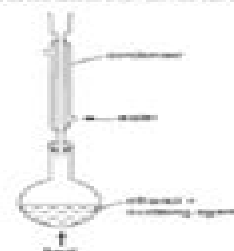
The list shows three chemical reactions.

- 1 combustion of ethanol
- 2 fermentation of glucose
- 3 reaction of ethanol with ethanoic acid to give an ester

In which reactions is water a product?

- A** 1 and 2 only **B** 1 and 3 only **C** 2 and 3 only **D** 1, 2 and 3

8. The reaction of ethanol to ethanoic acid often carried out in the apparatus shown.

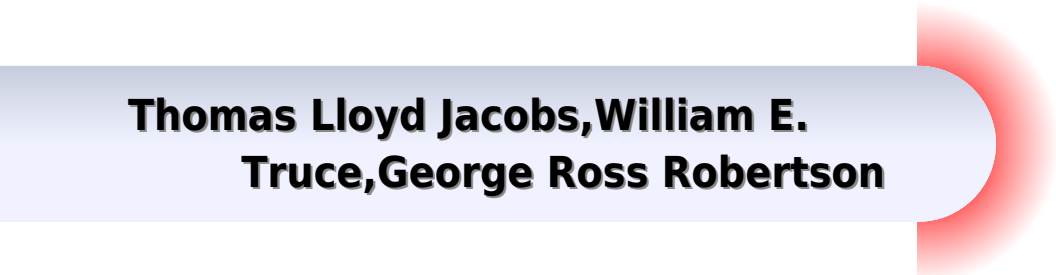


What is the purpose of the condenser?

- A** to prevent it reacting with the ethanoic acid
B to prevent any ethanol from escaping
C to prevent the ethanol acid changing back to ethanol
D to prevent the ethanoic acid reacting with the ethanol

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