

Background

- 1. Study of this _____ multi _____ relationship is a _____ method _____
- 2. Research on _____ balanced _____ equations _____

2. $4\text{Hg} + \text{Cl}_2 \rightarrow 2\text{Hg}_2\text{Cl}_2$

2. Magnesium reacting with a single oxygen atom produces 2 formula units of magnesium oxide.

The _____ is a _____ reaction _____ give the _____ ratio _____ for the _____ involved in the reaction _____

E.g. Problem

When aluminum aluminum reacts with elemental oxygen, aluminum oxide is produced



moles of _____



2. If you start with 4 moles of Al, how many moles of Al_2O_3 will be produced?

4 mol Al $\rightarrow$ 3 mol Al_2O_3 and $3\text{O}_2 \rightarrow 2$ mol Al_2O_3

Note Taking Guide Episode 801 Stoichiometry

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