

Note Taking Guide: Episode 301 Name_____

Model: A familiar idea used to explain unfamiliar facts observed in nature.

Theory: An explanation of observable facts and phenomena. To

remain valid, models and theories must:

- explain all known facts
- enable scientists to make correct predictions.

Democritus:

- proposed the existence of an atom
- word comes from the greek word atomis which means not to cut or "indivisible"

Aristotle

- rejected the idea of the atom
- said matter could be cut continually

Dalton's theory proposed that atoms:

- are the building blocks of matter
- are indivisible
- of the same elements are identical
- of different elements are different
- unite in small, whole number ratios to form compounds.

J.J. Thomson

- credited with discovery of electron: a blow to Dalton's indivisible atom.
- proposed the plum pudding model of the atom:
Negatively charged electrons embedded in a ball of positive charge

Note Taking Guide Episode 301 Chemistry

D Keegan



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In some sort of inundated with monitors and the cacophony of instantaneous communication, the profound energy and emotional resonance of verbal beauty frequently fade in to obscurity, eclipsed by the regular assault of noise and distractions. Yet, set within the musical pages of **Note Taking Guide Episode 301 Chemistry**, a fascinating perform of literary beauty that impulses with organic thoughts, lies an memorable journey waiting to be embarked upon. Written by way of a virtuoso wordsmith, that mesmerizing opus manuals visitors on a mental odyssey, softly revealing the latent potential and profound impact embedded within the delicate internet of language. Within the heart-wrenching expanse of the evocative analysis, we will embark upon an introspective exploration of the book is central styles, dissect its charming writing fashion, and immerse ourselves in the indelible effect it leaves upon the depths of readers souls.

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