

Nuclear Chemistry

- We hear about nuclear power and the results of the atom only when there is a Tragic accident

- Three Mile Island
- Chernobyl
- Fukushima

It is important to have a understanding of radiation and there effects

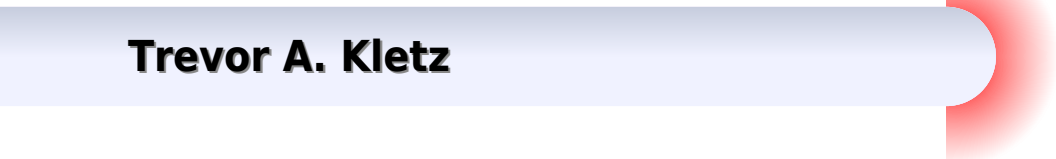
- Background Radiation
 - We are all exposed to radiation
 - Three-fourths of all exposure to radiation comes from **background radiation**.
 - Most of the remaining one-fourth comes from medical irradiation such as X-rays.
- Average origin of radiation we receive
- We receive about 0.3 rem of radiation each year
 - 55 % from radon in indoor and outdoor air
 - 8 % from cosmic rays from outer space
 - 8 % from rocks and soil
 - 11 % from natural radioactive isotopes (^{14}C) of elements that are present in our own bodies
 - 18 % from anthropogenic sources (mainly medical x- rays)
 - Contribution from nuclear power plants is negligible (on average)

Radon

- Guidelines for radon levels in indoor air:
 - Health Canada set the maximum to 200 Bq/m^3
 - U.S. EPA set the maximum to $4 \text{ pCi/L} = 148 \text{ Bq/m}^3$
- Radiation Damage to Cells
 - Radiation is capable of removing electrons from cells forming ions, hence the term **ionizing radiation**.
 - Molecules can also splinter into neutral fragments called **free radicals**. Free radicals can disrupt cellular processes.

Nuclear Chemistry Chem Guide

Trevor A. Kletz



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 - The Rise of Digital Reading Nuclear Chemistry Chem Guide
 - Advantages of eBooks Over Traditional Books
2. Identifying Nuclear Chemistry Chem Guide
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Nuclear Chemistry Chem Guide
 - User-Friendly Interface
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 - Personalized Recommendations
 - Nuclear Chemistry Chem Guide User Reviews and Ratings
 - Nuclear Chemistry Chem Guide and Bestseller Lists
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 - Interactive Elements Nuclear Chemistry Chem Guide
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 - Participating in Virtual Book Clubs
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 - Dealing with Digital Eye Strain
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 - Setting Reading Goals Nuclear Chemistry Chem Guide
 - Carving Out Dedicated Reading Time
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 - Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development

- Exploring Educational eBooks

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- Interactive and Gamified eBooks

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