

Nuclear Chemistry Practice Test

Multiple Choice: Identify the choice that best completes the statement or answers the question.

- ____ 1. In nuclear chemistry, an atom is referred to as a(n)
a. nuclide. b. nucleon. c. nucleus. d. alpha particle.
- ____ 2. What does the 101 in $^{256}_{101}\text{Md}$ represent?
a. the mass number b. the atomic number c. the nuclide number d. the number of neutrons
- ____ 3. The nuclear binding energy is released when a nucleus
a. is bombarded. c. is formed from its constituent particles.
b. divides. d. decays.
- ____ 4. Balance the following equation: $^{226}_{86}\text{Rn} \rightarrow ^{226}_{86}\text{Rn} + \underline{\hspace{1cm}}$
a. ^4_2He b. ^1_0n c. ^1_1H d. $^0_{-1}\text{e}$
- ____ 5. Balance the following equation: $^{238}_{92}\text{U} + \underline{\hspace{1cm}} \rightarrow ^{239}_{92}\text{U}$
a. ^4_2He b. ^1_0n c. ^1_1H d. $^0_{-1}\text{e}$
- ____ 6. Balance the following equation: $^9_4\text{Be} + ^4_2\text{He} \rightarrow ^{12}_6\text{C} + \underline{\hspace{1cm}}$
a. ^4_2He b. ^1_0n c. ^1_1H d. $^0_{-1}\text{e}$
- ____ 7. The spontaneous disintegration of a nucleus into a slightly lighter and more stable nucleus, accompanied by emission of particles, electromagnetic radiation, or both, is
a. nuclear fusion. b. nuclear radiation. c. nuclear decay. d. nuclear fission.
- ____ 8. During radioactive decay, the nucleus disintegrates into
a. a lighter and more stable nucleus. c. a lighter and less stable nucleus.
b. a heavier and more stable nucleus. d. a heavier and less stable nucleus.
- ____ 9. Which of the following radioactive decay processes does *not* reduce the atomic number of a nuclide?
a. alpha decay c. both alpha & beta decay
b. beta decay d. neither alpha nor beta decay
- ____ 10. Beta particles are
a. electrons. b. helium nuclei. c. pure energy. d. neutrons.
- ____ 11. Gamma rays are
a. electrons. b. helium nuclei. c. pure energy. d. neutrons.
- ____ 12. The half-life of an isotope is the time required for half the nuclei in a sample to undergo ____.
a. radioactive decay b. nuclear fission c. nuclear fusion d. chemical changes
- ____ 13. What is the half-life of an isotope if 125 g of a 500 g sample of the isotope remain after 3.0 years?
a. 1.5 years b. 2.5 years c. 3.5 years d. 4.5 years
- ____ 14. Which is *not* a parent nuclide?
a. uranium-238 b. lead-206 c. uranium-235 d. thorium-232
- ____ 15. A decay series ends with a ____.
a. stable nuclide. b. parent nuclide. c. fission reaction d. beta decay.
- ____ 16. Artificial radioactive nuclides are produced by
a. bombarding stable nuclei with particles.
b. using an accelerator to overcome the nuclear force.
c. beta emission.
d. fission of stable nuclides.
- ____ 17. How are elements artificially transmuted?
a. Stable nuclei are bombarded with charged particles.
b. Stable nuclei are bombarded with uncharged particles.
c. Stable nuclei are bombarded with charged and uncharged particles.
d. Unstable nuclei are bombarded with charged and uncharged particles.
- ____ 18. Which of the following has the greatest penetrating ability?
a. alpha particles b. beta particles c. gamma rays d. All are equal.
- ____ 19. What are often used to monitor the approximate radiation exposure of people working with radioactive materials?
a. film badges c. scintillation counters
b. X-ray films (radiographs) d. radioactive tracers

Nuclear Chemistry Practice Exams

Disha Experts

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thorough understanding of the concepts 4 Simple and Easy to Understand Language Solutions are presented in a straightforward and accessible manner *Partial Solutions Manual* Darrell D. Ebbing, David Bookin, 1998

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