

BUILD Vocabulary

ionic bond Chemical bond formed when one or more electrons are transferred from one atom to another.

ion atom that has a positive or negative charge.

covalent bond type of bond between atoms in which the electrons are shared.

molecule smallest unit of most compounds that displays all the properties of that compound.

van der Waals forces weak attraction that develops between oppositely charged regions of nearby molecules.

Preface The prefix *co-* and *co-* mean "together." A compound is made up of elements joined together, and a covalent bond connects atoms that share electrons together. **How are covalent bonds different from ionic bonds?**

Chemical Bonds

KEY QUESTION What are the main types of chemical bonds?

The atoms in compounds are held together by chemical bonds. Bonds are formed by the electrons in the outer shell, which are called valence electrons. The main types of chemical bonds are ionic bonds and covalent bonds.

Ionic Bonds **Ionic bonds** are formed when one or more electrons are transferred from one atom to another. This transfer causes the atom losing the electrons to become positively charged. The atom gaining the electrons becomes negatively charged. The positively and negatively charged atoms become **ions**. The oppositely charged ions are attracted to each other and form an ionic bond.

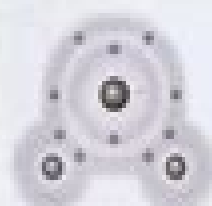
Covalent Bonds Sometimes atoms share electrons rather than transferring them. These shared electrons move around the nuclei of both atoms. The bond formed by sharing electrons is called a **covalent bond**. Atoms can share between two and six electrons. The structure that results is known as a **molecule**. A molecule is the smallest unit of most compounds.

Weak Interactions The strongest chemical bonds are ionic or covalent. Weak interactions between atoms are also important. Within living organisms, many molecules interact only briefly. **Van der Waals forces** are one type of weak interaction between molecules. They occur when molecules are very close together.

Hydrogen bonds are another type of weak interaction. They occur between a hydrogen atom of one molecule and an oxygen or nitrogen atom of a neighboring molecule.

Visual Reading Skill Covalent Bonds

1. What kind of chemical bond holds the water molecule together? _____
2. What particles are shared between the atoms in the molecule? _____
3. What is the other main type of chemical bond? _____
4. What is the difference between covalent and ionic bonds? _____



Water molecule (H_2O)

Properties of Water

READING TOOL Cause and Effect As you read the part of the lesson under "The Water Molecule," use the table below to list the causes and effects of the properties of water. Many of the causes may be the same, but each characteristic will have a different effect. Part of the table is completed for you.

Characteristic of Water	Cause	Effect
Hydrogen Bonding		The slightly negative end of a water molecule is attracted to the slightly positive end of another water molecule.
Expands Upon Freezing		
Dissolves Many Substances	Water forms hydrogen bonds.	
Cohesion		
Adhesion		
Heat Capacity		

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Scientific and Technical Books in Print, 1972 **RNA Isolation and Characterization Protocols** Ralph Rapley, David L. Manning, 1998-04-08 Ribonucleic acids are central to cellular and molecular processes and perform vital functions in both structural and functional roles RNA molecules form the bridge between the stable genetic information contained within DNA and enzymes and proteins that carry out much of the metabolism within the cell Many of the sites of protein synthesis the ribosomes within the cell are composed of these ribonucleic acids as are the tRNA molecules that deliver the amino acid building blocks to the ribosomes Of all the RNA species the nucleic acid intermediate messenger RNA is a desirable source of material to biologists since this reflects much of what ultimately is translated into enzymes and proteins In order to determine the qualitative and quantitative changes in mRNA expression a vast number of molecular biological techniques have been developed Key molecular methods that provide the means to initially isolate and analyze RNA molecules are the focus of this volume In putting together this collection of protocols we have tried to provide techniques that are most applicable and widely used In particular there are a number of isolation techniques included that have been developed modified or adapted to enable extraction from a variety of cell types organisms or subcellular organelles Successful isolation of intact RNA is an essential starting point for any subsequent analysis This is why we have aimed to make this section comprehensive The analysis of RNA is the focus of the following chapters *RNA Isolation and Analysis* Paul Jones, Jane Qiu, David Rickwood, 1994 The isolation and analysis of RNA plays a key role in molecular biology today This comprehensive account of the key techniques provides the background information and the protocols used to isolate and analyse RNA The book is essential reading for all researchers Molecular Biology of RNA Anna Robinson, 2023-09-19 Ribonucleic acid RNA is a nucleic acid which plays an integral role in coding decoding regulation and expression of genes Some of the different types of RNA are messenger RNA non coding RNA small chain RNA and long chain RNA Messenger RNA transmits the information from the DNA to the ribosome Non coding RNAs ncRNAs regulate the expression of genes at the transcriptional and post transcriptional level There are two major subtypes of regulatory non coding RNAs namely long non coding RNA lncRNA and micro RNA miRNA They both interact not only with each other but also with other intracellular components to modulate the inter related steps and mediators of regulated forms of cell death such as necrosis and apoptosis As a result they could be in a key position for regulating cell death This book provides significant information to help develop

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