

PHYSICAL CHEMISTRY

(Complete Syllabus)

SECTION - A

Straight Objective Type

This section contains 220 multiple choice questions. Each question has 4 choices (1), (2), (3) and (4), out of which **ONLY ONE** is correct.

Choose the correct answer :

- The mass of oxygen required to produce enough CO which completely reduces x gm of Fe_2O_3 . Suppose molecular weight of $\text{O}_2 = y$, molecular weight of $\text{Fe}_2\text{O}_3 = z$
(1) $\frac{2xz}{3y}$ (2) $\frac{3xy}{2z}$
(3) $\frac{2y}{3xz}$ (4) $\frac{3z}{2xy}$
- The number of lone pair of electrons present in 2.56 gm of orthorhombic sulphur will be
(1) 0.02 N_0 (2) 0.16 N_0
(3) 0.32 N_0 (4) 0.08 N_0
- A 10.0 g of sample of mixture of calcium chloride and sodium chloride is treated with Na_2CO_3 to precipitate the calcium as calcium carbonate. This CaCO_3 is heated to convert all calcium to CaO and final mass of CaO is 1.62 gms. The % by mass of CaCl_2 in the original mixture is
(1) 15.2%
(2) 32.1%
(3) 21.8%
(4) 11.07%
- 1000 ml of oxygen at STP were passed through an ozoniser and the resulting volume was 888 ml at STP. This quantity of ozonised oxygen is passed through excess of KI solution. The weight of I_2 liberated is
(1) 25.4 g (2) 12.7 g
(3) 2.54 g (4) 1.27 g
- x_1 ml solution having normality 1 is diluted to get a solution of normality y_2 . What would be volume required for dilution?
(1) $x_1 \left(1 - \frac{y_1}{y_2}\right)$ (2) $x_1 \left(\frac{y_2}{y_1} - 1\right)$
(3) $x_1 \left(\frac{y_1}{y_2} - 1\right)$ (4) $x_1 \left(1 + \frac{y_1}{y_2}\right)$
- 1.625 g of an interhalogen compound ICl_x was subjected to electric discharge which split it into constituents and brought to STP. It occupied 112 ml. The interhalogen compound is
(1) ICl (2) ICl_3
(3) ICl_5 (4) ICl_7
- If 12.25 g of a sample of KClO_3 of 75% purity and rest vapourisable impurities is strongly heated to give KCl , KClO_4 and 0.32 gm O_2 . The mass of residue obtained is
(1) 1.769 g (2) 7.096 g
(3) 5.327 g (4) 8.85 g
- A 2 g sample of xenon reacts with fluorine. The mass of compound produced is 3.158 g. The empirical formula of the compound is
(1) XeF_2 (2) XeF_4
(3) XeF_6 (4) XeF_8
- 1 g of carbonate of metal was dissolved in 25 ml of 1 N HCl . The resulting liquid required 5 ml of 1 N NaOH for neutralisation. The equivalent weight

Multiple Choice Physical Chemistry Exam Questions

Educart, Sir Tarun Rupani



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few months left for the examination it becomes quite challenging for students to cover all the concepts included in the NEET syllabus thoroughly. However, a proper study plan designed as per the latest examination pattern and the syllabus can help students to prepare all the important concepts in shorter time duration. Given below are few useful tips that can assist the students in tackling multiple choice questions in NEET exam accurately. In most of the multiple choice questions the options are designed in a very tricky and confusing manner. In most of the cases all the given options seem to be correct in some aspect. Therefore the students are advised to read the entire question very carefully. Try to accumulate all the information provided in the question effectively because in some of the cases you can easily evaluate the Answers from the question itself. If you are muddled by the given options then give each option a true and false test. Instead of getting confused consider all the possibilities and neglect the incorrect options. Hence in this way the most appropriate answer could be easily spotted.

Topic wise MCQs for NEET Chemistry: Environmental Chemistry, Organic Chemistry, P Block Elements, States of Matter, Redox Reaction, Chemical Equilibrium, Thermodynamics, Periodic Classification of Elements, s Block Elements, Chemical Bonding, Molecular Structure, Structure of Atom, Some Basic Concepts of Chemistry, Hydrocarbons, Some Basic Concepts of Chemistry, Solid State, Structure of Atom, Solutions, Classification of Elements and Periodicity in Properties, Electrochemistry, Chemical Bonding and Molecular Structure, Chemical Kinetics, States of Matter, Gases and Liquids, Surface Chemistry, Thermodynamics, General Principles and Processes of Isolation of Elements, Equilibrium, Redox Reactions, d and f Block Elements, Hydrogen, Coordination Compounds, Haloalkanes and Haloarenes, Some p Block Elements, Alcohols, Phenols and Ethers, Organic Chemistry, Some Basic Principles and Techniques, Aldehyde, Ketones and Carboxylic Acids, Hydrocarbons, Organic Compounds Containing Nitrogen, Environmental Chemistry, Biomolecules, Polymers and Chemistry in Everyday Life.

Use a step wise approach to solve conceptual and complex questions. Several times Matching type Questions are asked where the students are required to find the mismatched or the correctly matched option. Some of the questions asked in the NEET exam are entirely memory based therefore the students are advised to memorize the common names, scientific names, concepts and important definitions. Around 40% of the questions asked in the NEET exam are application based. Therefore students need to focus more on the concepts along with its applications in order to score well in the examination. Chemistry topics broadly can be grouped into Organic, Inorganic and Physical chemistry. Organic Chemistry is a vast section and holds the maximum weightage. Its concepts and principles are important. One needs to be conversant with properties of groups, conversions, recognizing organic compounds etc. Interlinking between different chapters from organic chemistry can be of great help. The students must primarily focus on reading NCERT textbooks. Several times the questions asked in NEET exam are taken directly from the NCERT textbooks. Initially avoid answering those questions for which you are not confident because your wrong answer may reduce your final score. In order to utilize your time appropriately divide the three hours of examination time as per your comfort among Physics, Chemistry and Biology. Initially focus on attempting all easy questions and later on

pick the difficult ones By this way your confidence will be elevated and you will also get more time to answer hard questions Practice previous years question papers mock tests and sample papers to get an idea on how to answer MCQ questions efficiently Preparing at an early stage is what an MCQ exam requires Avoid guesswork for negative marking questions as they might lower your final score These tips can be very helpful for students to answer difficult and brain teaser questions Prior preparations and practice are mandatory aspects of any examination Hence to crack highly competitive examination like NEET it is mandatory for students to prepare well and acquire the skills to tackle multiple choice questions effectively Rather than just following mere guesswork the aspirants can focus on the tips discussed to tackle Multiple Choice Questions in NEET in the right manner

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Papers Mathematics 1970 2022 with Value Added Notes covers the whole syllabus distributing in 24 Chapters The book comprises chapters such as Quadratic Equations and Expressions Complex Number Progressions Statistics and Probability Trigonometrically Ratios and Equations Differentiation Differential Equations Mathematical Reasoning and so on This book serves to be a suitable Study Guide for the aspirants with focus on Qualitative Preparation and Systematic understanding of the Syllabus and Examination Level With provision for self assessment in Mock Tests this book stands beneficial in imprinting concepts in the mind **The Pearson Guide To Physical Chemistry For The Aipmt** Singhal Atul,2011-09 **FCI**

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Levy,2022-02-13 The book describes growing up in NYC being the first member of his family to graduate from college and graduate school and describes the development of important new medical products and the formation of multiple early stage healthcare companies and the process of taking companies public and selling companies Competition Science Vision ,2005-01 Competition Science Vision monthly magazine is published by Pratiyogita Darpan Group in India and is one of the best Science monthly magazines available for medical entrance examination students in India Well qualified professionals of Physics Chemistry Zoology and Botany make contributions to this magazine and craft it with focus on providing complete and to the point study material for aspiring candidates The magazine covers General Knowledge Science and Technology news Interviews of toppers of examinations study material of Physics Chemistry Zoology and Botany with model papers reasoning test questions facts quiz contest general awareness and mental ability test in every monthly issue **Educart ICSE Class**

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