

06. The unit of K_c for reaction $N_2 + O_2 \rightleftharpoons 2NO$

- (a) mol dm^{-3} (b) $\text{mol}^{-2} \text{dm}^6$
(c) $\text{mol}^{-1} \text{dm}^3$ (d) ☒ no unit

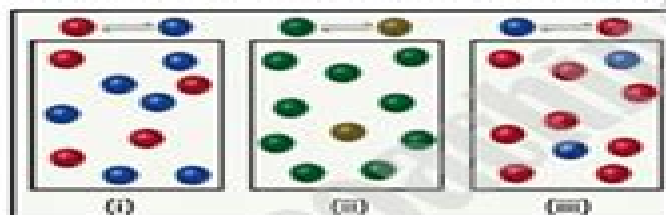
07. The system is stable in equilibrium when:

- (a) ☒ $Q_c = K_c$ (b) $Q_c > K_c$
(c) $Q_c < K_c$ (d) None of these

08. Q_c can be defined as:

- (a) ratio of product and reactants
(b) ☒ ratio of molar concentration of product and reactant at specific time
(c) ratio of molar concentration of product and molar volume of reactant
(d) ratio of molar concentration of product and reactant raised to the power of coefficient

09. Which of the following represent backward reaction?



- (a) (i) and (ii) (b) (ii) and (iii)
(c) (ii) only (d) ☒ (iii) only

10. The value of K_c increases when:

- (a) [Product] is less (b) ☒ [Product] is more
(c) [Reactant] is more (d) [Reactant = product]

11. Which of these are reversible reactions?

- (a) Condensation (b) Evaporation
(c) ☒ Condensation & Evaporation (d) None of them

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Terry C. Jones



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1. Understanding the eBook Multiple Choice Questions And Answers In Chemical Equilibrium
 - The Rise of Digital Reading Multiple Choice Questions And Answers In Chemical Equilibrium
 - Advantages of eBooks Over Traditional Books
2. Identifying Multiple Choice Questions And Answers In Chemical Equilibrium
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Multiple Choice Questions And Answers In Chemical Equilibrium
 - User-Friendly Interface
4. Exploring eBook Recommendations from Multiple Choice Questions And Answers In Chemical Equilibrium
 - Personalized Recommendations
 - Multiple Choice Questions And Answers In Chemical Equilibrium User Reviews and Ratings
 - Multiple Choice Questions And Answers In Chemical Equilibrium and Bestseller Lists
5. Accessing Multiple Choice Questions And Answers In Chemical Equilibrium Free and Paid eBooks
 - Multiple Choice Questions And Answers In Chemical Equilibrium Public Domain eBooks
 - Multiple Choice Questions And Answers In Chemical Equilibrium eBook Subscription Services
 - Multiple Choice Questions And Answers In Chemical Equilibrium Budget-Friendly Options
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 - Multiple Choice Questions And Answers In Chemical Equilibrium Enhanced eBook Features
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 - Highlighting and Note-Taking Multiple Choice Questions And Answers In Chemical Equilibrium
 - Interactive Elements Multiple Choice Questions And Answers In Chemical Equilibrium

8. Staying Engaged with Multiple Choice Questions And Answers In Chemical Equilibrium
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Multiple Choice Questions And Answers In Chemical Equilibrium
9. Balancing eBooks and Physical Books Multiple Choice Questions And Answers In Chemical Equilibrium
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Multiple Choice Questions And Answers In Chemical Equilibrium
10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Multiple Choice Questions And Answers In Chemical Equilibrium
 - Setting Reading Goals Multiple Choice Questions And Answers In Chemical Equilibrium
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Multiple Choice Questions And Answers In Chemical Equilibrium
 - Fact-Checking eBook Content of Multiple Choice Questions And Answers In Chemical Equilibrium
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
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 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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