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- (D) None of these
- The number of tetrahedral and octahedral voids in bexagonal prismatic unit cell HCP is
  - (A) 8.4 (C)

- (BO) 2.1
- 12.6
- (D) 6,12
- Which of the following void is smallest in close 16. packing?
  - (A) Tetrahedral
- (BA) Octabedral.
- Cubic (C)
- (D) Source
- 17. The number of octahedral void in BCC structure is
  - (A) = 0

(B)

(C) 2

- (D) 4
- Which of the following interstitial site is 18. formed when the three closed packed spheres of one layer is put over three closed packed spheres of the second layer, their positions being inverted with respect to each other? (B) Octoberdial
  - (A) Tetrahedral (C) Rhombohedral
    - (D) Tetragonal
- 19. In which crystal, alternate tetrahedral voids are occupied?
  - (A) NaCl (C) ZnS

- (B) ColF<sub>2</sub> (D) NasO
- A metallic element crystallizes into a lattice 250. containing a sequence of layers of ABABAB. Any packing of spheres leaves out voids in the lattice. Approximately what percentage by volume of this lattice is empty space?
  - (A) 74%

26% CES)

(C) 48%

- (D) 32%
- Packing efficiency of an identical solid sphere is 74% in
  - (A) Simple cubic structure
  - (B) Monoclinic structure
  - (C) HCP structure
  - (D) BCC structure

- Which of the following expression represents 22. total volume of atoms present in a FCC unit cell of a metal? (r is the atomic radius)

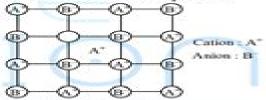
(C)  $\frac{12}{2}\pi r^3$ 

- 23. In a BCC packing, the nearest neighbours lie along the
  - (A) edges of the cube
  - fisce diagonal.
  - line joining the two opposite corners of (C) the face.
  - cube diagonal (ID)
- 24. The ionic radii of Rb<sup>+</sup> and I<sup>-</sup> are 1.46 and 2.16

A, respectively. The most probable type of structure exhibited by it is

- (A) CsCl type
- (HS) NaCl type:
- ZnS type (C)
- (D) CaF<sub>2</sub> type
- 25. The radius of Ag' ion is 126 pm while of I ion is 216 pm. The coordination number of Ag in Agil is
  - (A) 2 (C)

- (IBA): (D) 8
- 26. Structure shown here represents:



- Schottky defect CALL
- Frenkel defect CESS
- CCD Metal excess defect
- None of these (ED)
- 27. Select the incorrect statement:
  - (A) Stoichiometry of crystal uneffected due to Schottky defect.

# **Multiple Choice Questions Of Solid State Chemistry**

**Peipei Pang** 

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