

Oxidation Numbers

Summary: An atom connected to a **more** eN atom = +1

An atom connected to a **less** eN atom = -1

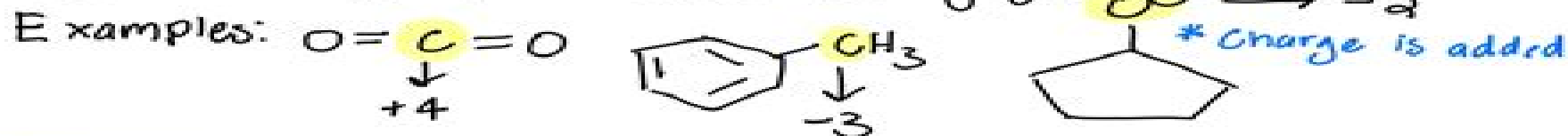
An atom connected the **same** atom = 0

* Number of bonds must be taken into account

* Charge is also incorporated into oxidation number

Oxidation number $\uparrow$ = Oxidized = reducing agent

Oxidation number $\downarrow$ = reduced = oxidizing agent



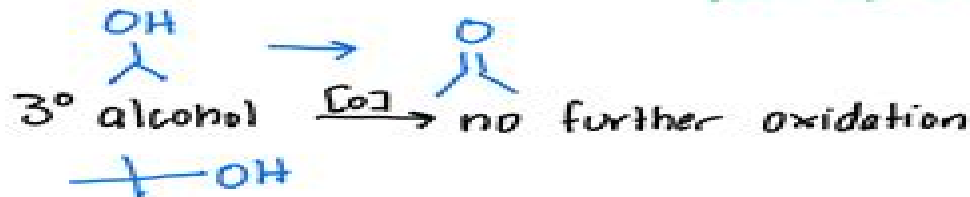
Oxidation Reactions

Summary: 1° alcohol $\xrightarrow{[\text{O}]}$ aldehyde $\xrightarrow[\text{H}_2\text{O}]{\text{must be wet } [\text{O}]}$ carboxylic acid

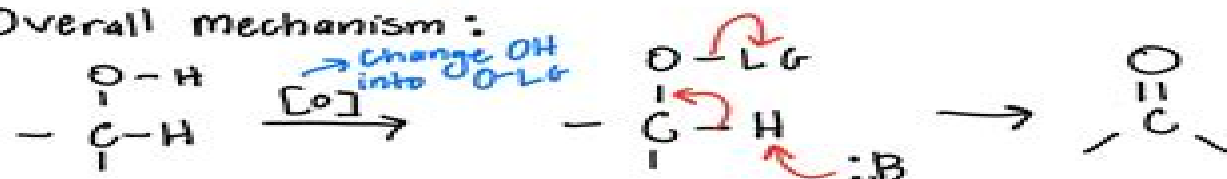


1° alcohol $\xrightarrow{[\text{O}]}$ aldehyde must be dry

2° alcohol $\xrightarrow{[\text{O}]}$ ketone can be either wet or dry



Overall mechanism:



Organic Chem Phenol Study Guide

Mr. Rohit Manglik



Organic Chem Phenol Study Guide:

A Self-study Guide to the Principles of Organic Chemistry Jiben Roy, 2013 A Self Study Guide to the Principles of Organic Chemistry Key Concepts Reaction Mechanisms and Practice Questions for the Beginner will help students new to organic chemistry grasp the key concepts of the subject quickly and easily as well as build a strong foundation for future study Starting with the definition of atom the author explains molecules electronic configuration bonding hydrocarbons polar reaction mechanisms stereochemistry reaction varieties organic spectroscopy aromaticity and aromatic reactions biomolecules organic polymers and a synthetic approach to organic compounds The over one hundred diagrams and charts contained in this volume will help students visualize the structures and bonds as they read the text and make the logic of organic chemistry clear and easily understood Each chapter ends with a list of frequently asked questions and answers followed by additional practice problems Answers are included in the Appendix

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Computational Approaches for the Prediction of pKa Values describes the insights that have been gained on the intrinsic and extrinsic features that influence a molecule's acidity and discusses the computational methods developed to estimate acidity from a compound's molecular structure. The authors examine the strengths and weaknesses of the theoretical techniques and show how they have been used to obtain information about the acidities of different classes of chemical compounds. The book presents theoretical methods for both general and more specific applications covering methods for various acids in aqueous solutions including oxyacids and related compounds, nitrogen acids, inorganic acids, and excited state acids, as well as acids in nonaqueous solvents. It also considers temperature effects, isotope effects, and other important factors that influence pKa. This book provides a resource for predicting pKa values and understanding the bases for these determinations, which can be helpful in designing better chemicals for future uses.

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Table of Contents Organic Chem Phenol Study Guide

1. Understanding the eBook Organic Chem Phenol Study Guide
 - The Rise of Digital Reading Organic Chem Phenol Study Guide
 - Advantages of eBooks Over Traditional Books
2. Identifying Organic Chem Phenol Study Guide
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Organic Chem Phenol Study Guide
 - User-Friendly Interface
4. Exploring eBook Recommendations from Organic Chem Phenol Study Guide
 - Personalized Recommendations
 - Organic Chem Phenol Study Guide User Reviews and Ratings
 - Organic Chem Phenol Study Guide and Bestseller Lists
5. Accessing Organic Chem Phenol Study Guide Free and Paid eBooks

- Organic Chem Phenol Study Guide Public Domain eBooks
- Organic Chem Phenol Study Guide eBook Subscription Services
- Organic Chem Phenol Study Guide Budget-Friendly Options
- 6. Navigating Organic Chem Phenol Study Guide eBook Formats
 - ePub, PDF, MOBI, and More
 - Organic Chem Phenol Study Guide Compatibility with Devices
 - Organic Chem Phenol Study Guide Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Organic Chem Phenol Study Guide
 - Highlighting and Note-Taking Organic Chem Phenol Study Guide
 - Interactive Elements Organic Chem Phenol Study Guide
- 8. Staying Engaged with Organic Chem Phenol Study Guide
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Organic Chem Phenol Study Guide
- 9. Balancing eBooks and Physical Books Organic Chem Phenol Study Guide
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Organic Chem Phenol Study Guide
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Organic Chem Phenol Study Guide
 - Setting Reading Goals Organic Chem Phenol Study Guide
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Organic Chem Phenol Study Guide
 - Fact-Checking eBook Content of Organic Chem Phenol Study Guide
 - Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development

- Exploring Educational eBooks

14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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