

# YOUR NAME

PRODUCT DESIGNER, ANALYTICS



## PROFILE

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🏠 New York, USA  
☎ +01 2345 6789 02  
✉ jacobscott@email.com

## SKILLS

### PROFESSIONAL

- Graphic Design
- UX/UI Design
- Web Design
- Project Management
- Market Research
- Wireframing
- Usability Testing

## EDUCATION

- **DEGREE TITLE**  
University Name  
2020 - 2023
- **DEGREE TITLE**  
University Name  
2020 - 2023

## EXPERTISE

### TECHNICAL

- Adobe Photoshop
- Adobe Illustrator
- After Effects
- Sketch App

## EXPERIENCE

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• 2020 -

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**William Van Hecke**



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web he concluded that at a constant pressure the volume of the gas would expand at a particular rate for each degree of

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web the volume of 1 00 mol 1 00 mol of any gas at stp standard temperature 273 15 k and pressure 1 atm is measured to be 22 414l 22 414 l we can substitute 101 325kpa 101 325 kpa for pressure 22 414 l 22 414 l for volume and 273 15 k 273 15 k for temperature into the ideal gas equation and solve for r r

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web chemists sometimes make comparisons against a standard temperature and pressure stp for reporting properties of gases 273 15 k and 1 atm 101 325 kpa 1 at stp one mole of an ideal gas has a volume of about 22 4 l this is referred to as the standard molar volume figure 9 18

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web pressure increases even though there were errors in the experiment the goal of the experiment was met the goal of declaring what type of relationship it is was clear using the data pressure kpa temperature c temperature k constant k p t 95 82 kpa 16 0 c 289 k 0 33 97 90 kpa 19 0 c 292 k 0 34 98 97 kpa 99 22 kpa 101 4 kpa 19 1 c

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web the properties of water at constant volume can be represented as a function of pressure and temperature as shown in the diagram below figure 1 the pressure temperature relationship at constant volume the saturation point of water is the condition at which a phase change occurs from liquid to vapour or vapour to liquid

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