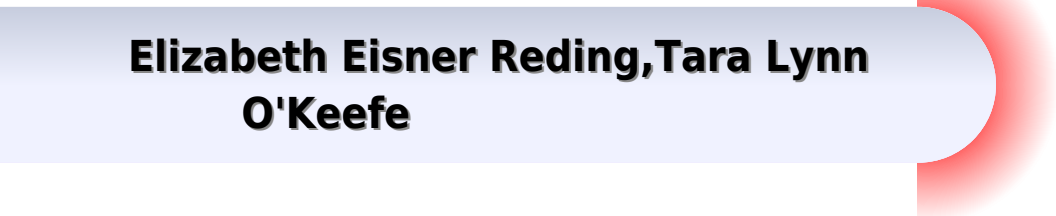




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**Elizabeth Eisner Reding, Tara Lynn
O'Keefe**



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web the mass of the given substance is converted into moles by use of the molar mass of that substance from the periodic table then the moles of the given substance are converted

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web how many moles of salt are in 13.8 g NaCl text g 13.8 g 13 point 8 start text g end text of sodium chloride express the answer using 3 significant figures your answer should be

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web collectively these conversions are called mole mass calculations a stoichiometry calculation converting between masses and moles of different substances in a chemical

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web sep 21 2022 while the mole ratio is ever present in all stoichiometry calculations amounts of substances in the laboratory are most often measured by mass therefore