

dengan $\gamma = \frac{C_p}{C_v}$ = konstanta Laplace, dan $\frac{C_p}{C_v} > 1$. C_p adalah kapasitas kalor gas pada tekanan tetap dan C_v adalah kalor gas pada volume tetap. Perhatikan diagram $p - V$ pada Gambar 9.7. Dari kurva hubungan $p - V$ tersebut, Anda dapat mengetahui bahwa:

- 1) Kurva proses adiabatik lebih curam daripada kurva proses isothermal.
- 2) Sama, tekanan, merupakan volume pada proses adiabatik tidak tetap.

Oleh karena sistem tidak melepaskan atau menerima kalor, pada kalor sistem proses adiabatik Q sama dengan nol. Dengan demikian, usaha yang dilakukan oleh sistem hanya mengulangi energi dalam sistem tersebut. Biasanya usaha pada proses adiabatik tersebut dinyatakan dengan persamaan berikut.

$$W = \frac{3}{2} nR(T_1 - T_2) = \frac{5}{2} (p_1 V_1 - p_2 V_2) \quad (9-8)$$

Contoh 9.6

Sebuah mesin memiliki rasio pemampatan $12 : 1$ yang berarti bahwa setelah pemampatan, volume gas menjadi $\frac{1}{12}$ volume awalnya. Anggap bahan bakar bercampur udara pada suhu 33°C , tekanan 1 atm, dan $\gamma = 1,4$. Jika proses pemampatan terjadi secara adiabatik, hitunglah tekanan pada keadaan akhir dan suhu campuran.

Jawab:

Diketahui: $V_2 = \frac{1}{12} V_1$, $T_1 = 33 + 273 = 306 \text{ K}$, dan $p_1 = 1 \text{ atm}$.

Untuk menentukan tekanan akhir p_2 , gunakan rumus:

$$p_1 V_1^\gamma = p_2 V_2^\gamma \rightarrow p_2 = p_1 \left(\frac{V_1}{V_2} \right)^\gamma \rightarrow p_2 = 1 \left(\frac{V_1}{\frac{1}{12} V_1} \right)^{1,4} = (12)^{1,4} = 32,4 \text{ atm.}$$

Suhu campuran akan akhir akhir T_2 diperoleh sebagai berikut:

$$T_1 V_1^{\gamma-1} = T_2 V_2^{\gamma-1} \rightarrow T_2 = T_1 \left(\frac{V_1}{V_2} \right)^{\gamma-1}$$

$$T_2 = 306 \text{ K} (12)^{1,4-1} = 306 \text{ K} (12)^{0,4} = 532 \text{ K} = 259^\circ\text{C}$$

Contoh 9.7

Udara sebesar $2 \times 10^3 \text{ J}$ diberikan secara adiabatik untuk memampatkan $0,5 \text{ mol}$ gas ideal monatomik, sehingga suhu mutlaknya menjadi 2 kali semula. Jika konstanta umum gas $R = 8,31 \text{ J/mol K}$, tentukanlah suhu awal gas.

Jawab:

Diketahui: $W = 2 \times 10^3 \text{ J}$, $T_2 = 2T_1$ dan $n = 0,5 \text{ mol}$.

$$W = \frac{3}{2} nR(T_2 - T_1) = \frac{3}{2} nR(2T_1 - T_1)$$

$$= \frac{3}{2} nR T_1$$

$$T_1 = \frac{2W}{3nR} = \frac{2(2 \times 10^3 \text{ joule})}{3 \times 0,5 \text{ mol} \times 8,31 \text{ J/molK}} = 321 \text{ K}$$

Jadi, suhu awal gas adalah 321 K.



Gambar 9.7

Pada proses adiabatik, kurva $p - V$ lebih curam dibanding kurva dengan kurva $p - V$ pada proses isothermal.

Lompatan Kuantum



1 ... 2 atom thermal energy, absorbed when light (photons) have high freq. When an atom's temperature perturbed, atom emits perturbed light yang terang dan berenergi high yang dingin. Proses pertubuhan ini is lompatan kuantum.

Quantum

Leap

the atom's thermal energy, absorbed when light (photons) have high freq. When an atom's temperature is perturbed, atom emits perturbed light yang terang dan berenergi high yang dingin. Proses pertubuhan ini is a jumping process (quantum leap).

Source: conceptual physics, 12th ed.

Conceptual Physics Fundamentals Answers

**Michael Matthews, Colin F.
Gauld, Arthur Stinner**



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Fundamental Physics and Physics Education Research Burra G. Sidharth, Jesús Carnicer Murillo, Marisa Michelini, Carmen Perea, 2020-11-11 This book highlights selected contributions presented at the 15th annual international symposium Frontiers of Fundamental Physics FFP15 with the aim of informing readers about the most important recent advances in fundamental physics and physics education research The FFP series offers a platform for physicists from around the world to present their latest theories and findings The latest symposium was held in Orihuela Spain and covered diverse fields of research including gravitation astronomy and astrophysics physics of complex systems high energy physics and mathematical physics Considerable attention was also paid to physics education research teacher education in physics and the popularization of physics In a knowledge based society research into fundamental physics plays a vital role in both the advancement of human knowledge and the development of new technologies Presenting valuable new peer reviewed contributions submitted from 15 countries this book will appeal to a broad audience of scholars and researchers

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Table of Contents Onceptual Hysics Undamentals Nswers

1. Understanding the eBook Onceptual Hysics Undamentals Nswers
 - The Rise of Digital Reading Onceptual Hysics Undamentals Nswers
 - Advantages of eBooks Over Traditional Books
2. Identifying Onceptual Hysics Undamentals Nswers
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Onceptual Hysics Undamentals Nswers
 - User-Friendly Interface
4. Exploring eBook Recommendations from Onceptual Hysics Undamentals Nswers
 - Personalized Recommendations
 - Onceptual Hysics Undamentals Nswers User Reviews and Ratings
 - Onceptual Hysics Undamentals Nswers and Bestseller Lists

5. Accessing Onceptual Hysics Undamentals Nswers Free and Paid eBooks
 - Onceptual Hysics Undamentals Nswers Public Domain eBooks
 - Onceptual Hysics Undamentals Nswers eBook Subscription Services
 - Onceptual Hysics Undamentals Nswers Budget-Friendly Options
6. Navigating Onceptual Hysics Undamentals Nswers eBook Formats
 - ePub, PDF, MOBI, and More
 - Onceptual Hysics Undamentals Nswers Compatibility with Devices
 - Onceptual Hysics Undamentals Nswers Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Onceptual Hysics Undamentals Nswers
 - Highlighting and Note-Taking Onceptual Hysics Undamentals Nswers
 - Interactive Elements Onceptual Hysics Undamentals Nswers
8. Staying Engaged with Onceptual Hysics Undamentals Nswers
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Onceptual Hysics Undamentals Nswers
9. Balancing eBooks and Physical Books Onceptual Hysics Undamentals Nswers
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Onceptual Hysics Undamentals Nswers
10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Onceptual Hysics Undamentals Nswers
 - Setting Reading Goals Onceptual Hysics Undamentals Nswers
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Onceptual Hysics Undamentals Nswers
 - Fact-Checking eBook Content of Onceptual Hysics Undamentals Nswers
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning

- Utilizing eBooks for Skill Development
- Exploring Educational eBooks

14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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