ORGANIC CHEMISTRY FUNDAMENTALS

Types of Organic Compounds

Organic chemistry is the study of natural and synthetic materials that here curbon atoms as the key chemical feature. There are more than one million known organic compounds.

HYDROCARBON

- ALKANE C-H
- Methyl (Me): -CH,
 Ethyl (Et): -C,H,

ALKENE

- Etheres C.H.
 Dienes Two C+C
- Tirlemer Three C+C

AUCYNE

Ethysies C.H.

-0=0-

AROMATIC

- Benzene: C_iH_e
 Arene: C_iH_e (Ar-)
- -O- ADDED
- ALCOHOL Methanol: Me-OH
- (methyl alcohol)
 Phonol: Ar-OH
- Diet/Glycel: (2 -OH)
 Gborrel: (3 -OH)
- ETHER
- Et-O-Et (diefre) ether)
- EPOXY
- Onlines
- C_H_O (ethylene oxide)
 PEROXIDE R
- PEROXIDE R-O-O-R'

 Aceteer Peroxide: C.H.O.

Benzuyl Perunlile: C. H. O.

>C =O ADDED

R-OH.

- · Methanal: H,CO (formaldeleade)
- Benzahbehoder Ar-CHO

KETONE NET

- 2-peopassee: Mo-CO-Me (dimethyl ketone, acetone)
- Dillustomes R-CO-R*-CO-R*

>COO ADDED

- Ethanole Arid: Me-COOH (sectic sold)
- Acctude Bear Me-COO:
- Benneic Acids As-COOHI

Dicartesylic Acid HOOC-R-COOH

OSTOR

- Ethyl Acetate:
 Me-CO-OEth
- Other derivatives:
- Persoyacids R-CO-COH.
- Add Antoholde: RCO-O-COR:

NITROGEN ADDED AMINE

- N R-N-R
- Methyl Amine: H₂C-NII₂
- Phenylamine: Ar-Nill, (aniline)
- * R-NH, (1'), RR'NH (2'), RR'R'N (3')
- NITRO R-NO.
- DIAZO R-NEN
- NITRILE R-CEN

* Mothone Nitrite: Mc-CN

AMEDIC

Acctionable: Me-CO-NEL



R-XI

SULFUR ADDED

- Thiotic R-SHI
- Thiorethorn R-S-R*
- Dismillion; R.S.S.R.
- Thiol Ester: R-CO-SR*
- Sulfinable: R-SO-R*
- Sufficient R-SO₂-R*
- Sulfinnic Artids R-SO III

HALOGEN ADDED

- Habatkaner
- Me-Cl eldoromethone
- Halobensenet Ar-X
- Chlorobenzone: Ar-Cl
- Acyl Ballider R-CO-X
- Aryli Hallide: Ar-X

Formulas & Isomers

- Molecular Formula: Elemental symbols with subscripts denote the composition of a compound
 - Empirical Fermula: Subscripts denote the relative elemental composition
 - Cimphical depiction
 - Dash Formula: Diagram all atoms; show bonds to dashes
 - Bond-Line Formula: Hide II; curbon atoms are depicted as lines, but other stores are shown explicitly



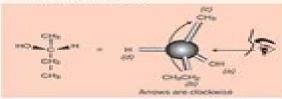
- Newman Projections 2-d depiction
- 3-dimensionals Wedges of sanshorse denote structure

SUBDIVISIONS OF ISOMERS

Constitutional loaners Storesinomers
Atoms have a Same connecting building safety at the amangation of the same at the same at

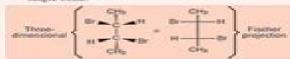
- Constitutional Isomers: Different bonding connectivity (e.g., rings, bonds, branching, substituent positions)
- Tautomers: Easily interconverted structural isomers (e.c., kete-end for ketone)
- · Chiral: Not identical with its mirror image
- · Achieut: Has a plane of symmetry (i.e.,
- superimposable on its mirror image)

 Epimerro: Pair of disstancements which differ only in
 the configuration of one atom
- . More than I chiral center
 - n chiral centers, ≤ 2º stereosomers
- Meso: 2 chiral centers, 4 isomers: 3 stereoisomers, 1 achiral (mirror-plane)

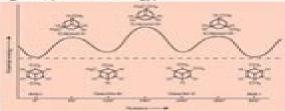


- R/S. Nutation: The 4 different atoms or groups attached to a central atom are ranked (a), (b), (c), and (d), by molar mass. The lowest (d) is directed away from the viewer and the sequence of a-b-c produces clockwise (R) or counter-clockwise (S) configuration.
 - Chiral (Optically Artive)c + or the rotation of plane polarized light. R/S: opposite effects

- Recentle: 50:50 mixture of stereoisomers (no net optical activity)
- Numerorlature: Note R/S and +/- in the compound name (c.z., R (+) bronnochloromethanel)
- Fisher-Projection: Diagram depicts chiral@-D
 structure
 - Molecular Conformations: Molecule exhibits structural variation due to free rotation about C-C single bond;



 Newman-Blagrame Depict rotation about a C-C bond; orlipsed (high crocqy), anti (less energy), and genethe (intermediate energy)



Common Terms

Allphastic: Non-aromatic Aromatic: Benzone ring

ion formation

Conjugation: Sequence of alternating double (or triple) and single bonds Dielectric Effect: Polar solvent stabilizes

Exergonic Reaction: Chemical reaction

that loses energy during the process of the reaction Heterocyclic: Non-carbon atom in the

ring structure
Hydrocarbon: Compound of II and C
Ketese: Molocule made up of a

enonovace/haride bonded to a lectone

Monasaccharider Carbohydrate that cannot be reduced by hydrolysis into another simple sugar Obelia: Alkona

Paraffine Alkane Radioisotope: Isotope that is radioactive

arbohydrate that | Saturated: Monimum # of Hs (all C-C single bonds) | Tetrese: Monosaccharide with four

carbon atoms
Unsaturated: At least one C-C multiple bond
Zymangy: The study of the process of
ferminatation

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Subhask, Sikdar, 2017-11-22 FROM THE INTRODUCTION This three volume set Bioremediation Principles and Practice provides state of the art description of advances in pollution treatment and reduction using biological means identify and address at a fundamental level broad scientific and technological areas that are unique to the subject or theme and that must be understood if advances are to be made and provide a comprehensive overview of new developments at the regulatory desk top bench scale pilot scale and full scale levels The set covers all media air water and soil sediment and blends the talents knowledge and know how of academic industrial governmental and international contributors The set addresses the removal of both hazardous and nonhazardous contaminants from the liquid solid and gas phase using biological processes This includes the biological treatment of wastes of municipal and industrial origin bioremediation of leachates soils and sediments and biofiltration for contaminated gases **Organic Chemistry** John M. McIntosh, 2018-03-05 Written in a concise and student friendly way this textbook focuses on the underlying principles of organic chemistry and provides the tools for understanding the nature of organic reactions The author utilizes an integrated approach for organic chemistry uniting in a logical manner the main reaction types and their mechanisms compound classes and their typical reactions organic spectroscopy and principles of structure elucidation Organic Chemistry Fundamentals Melody Frost, 2025-08-25 Organic chemistry is the branch of chemistry that focuses on the study of carbon containing compounds particularly those that also contain hydrogen These compounds known as organic compounds are diverse and can be categorized into various classes such as functional groups aliphatic compounds aromatic compounds heterocyclic compounds and polymers Organic compounds form the foundation of all life on Earth and make up the majority of known chemicals including agrichemicals petrochemicals and pharmaceuticals This discipline examines the properties structures and reactions of these compounds examining both qualitative and quantitative characteristics such as melting and boiling points conductivity solubility refractive index odor and color Analytical techniques like gas chromatography high performance liquid chromatography HPLC and separation methods such as crystallization distillation and solvent extraction are frequently used in organic chemistry to assess purity and isolate compounds The topics included in this book on organic chemistry are of utmost significance and bound to provide incredible insights to readers It explores all the important aspects of the field in the present day scenario It will serve as a valuable source of reference for those interested in this field **Fundamentals of Organic Chemistry** Monson R. S., 1985 Fundamentals of Organic Chemistry Monson R. S., 1985 **Fundamentals** of Organic Chemistry ,1994 Fundamentals of Organic Chemistry T. W. Graham Solomons, 1998-02-03 A realistic approach to the study of mechanisms The book addresses real functional group chemistry with an emphasis on the biological environmental and medical applications of organic chemistry **Fundamentals of Organic Chemistry + Fundamentals** of Organic Chemistry Study Guide + Organic and Inorganic Chemistry Molecular Set Mcmurry, 2006-09-01 Fundamentals of Organic Chemistry, Study Guide T. W. Graham Solomons, 1990-01-16 The Third Edition of the

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