

ORGANIC CHEMISTRY

REACTIONS

Reaction of alcohols with sodium

Alcohols react with sodium to form alkoxides and hydrogen gas.

$$2\text{C}_2\text{H}_5\text{OH} + 2\text{Na} \rightarrow 2\text{C}_2\text{H}_5\text{ONa} + \text{H}_2$$

The reaction is exothermic and vigorous, especially with primary alcohols.

Reaction of alcohols with phosphorus pentachloride

Alcohols react with PCl_5 to form alkyl chlorides and phosphorus oxychloride.

$$\text{C}_2\text{H}_5\text{OH} + \text{PCl}_5 \rightarrow \text{C}_2\text{H}_5\text{Cl} + \text{POCl}_3$$

The reaction is exothermic and produces a white solid.

Reaction of alcohols with acetic anhydride

Alcohols react with acetic anhydride to form esters and acetic acid.

$$\text{C}_2\text{H}_5\text{OH} + (\text{CH}_3\text{CO})_2\text{O} \rightarrow \text{CH}_3\text{COOC}_2\text{H}_5 + \text{CH}_3\text{COOH}$$

The reaction is exothermic and produces a pleasant odor.

Reaction of alcohols with phosphorus trichloride

Alcohols react with PCl_3 to form alkyl chlorides and phosphorus trichloride.

$$3\text{C}_2\text{H}_5\text{OH} + \text{PCl}_3 \rightarrow 3\text{C}_2\text{H}_5\text{Cl} + \text{P(O)Cl}_3$$

The reaction is exothermic and produces a white solid.

Reaction of alcohols with sodium metal

Alcohols react with sodium metal to form alkoxides and hydrogen gas.

$$2\text{C}_2\text{H}_5\text{OH} + 2\text{Na} \rightarrow 2\text{C}_2\text{H}_5\text{ONa} + \text{H}_2$$

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Reaction of alcohols with sodium metal

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The reaction is exothermic and vigorous, especially with primary alcohols.

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