

ORGANIC COMPOUNDS STUDY GUIDE

Remember that you should be using strategy



Name: _____ Date: _____

Organic Compound

Function

1. _____
 Name: _____ and provide structural support to plants.
 Function: _____ (long-term energy storage)
2. _____
 Name: _____ provides some cooling power.
 Function: _____ (cooling liquid, heat up, evaporate, and release)
3. _____
 Name: _____ provides structural support in plants.
 Function: _____ (structural support, water transport, and photosynthesis)

4. Nucleic Acids

4. Nucleic acids store genetic information. They are made of repeating units of organic compounds called nucleotides. Nucleotides are made of a sugar, a phosphate group, and a nitrogenous base.



NUCLEIC ACIDS

4a. The structure of a nucleic acid is a repeating chain of the diagram above.



- 4b. Nucleotides are the repeating building blocks of a larger molecule called a nucleic acid. Information is stored in the _____.
- 4c. _____ is a single nucleotide that is the repeating unit of a nucleic acid.
- 4d. What are nucleic acids used for? (What do they do?)

PROTEINS

4e. Proteins are made up of amino acids. Amino acids are the building blocks of proteins. There are 20 different amino acids.

4f. Each amino acid has a central carbon atom bonded to a hydrogen atom, an amino group, a carboxyl group, and a variable group (R group). The variable group is what makes each amino acid unique.

- 4g. _____ is a short chain of amino acids.
- 4h. _____ is a long chain of amino acids.
- 4i. _____ is a short chain of amino acids.
- 4j. _____ is a long chain of amino acids.
- 4k. _____ is a short chain of amino acids.
- 4l. _____ is a long chain of amino acids.
- 4m. _____ is a short chain of amino acids.
- 4n. _____ is a long chain of amino acids.
- 4o. _____ is a short chain of amino acids.
- 4p. _____ is a long chain of amino acids.
- 4q. _____ is a short chain of amino acids.
- 4r. _____ is a long chain of amino acids.

ORGANIC COMPOUNDS STUDY GUIDE

4a. Remember that you should be using strategy

Organic Compound	POLYMER	Can be broken down	MONOMER
1. Carbohydrates		Yes	
2. Nucleic acids		Yes	
3. Proteins		Yes	

4b. The organic compounds in food

CARBOHYDRATES

- 4c. Describe the chemical composition of carbohydrates. (What elements do they contain?)
- 4d. Carbohydrates are sugars. (What are examples of sugars?)
- 4e. Name 3 carbohydrates that you eat every day.
- 4f. What is the difference in the chemical structure of monomers and polymers?

PROTEINS

4g. Remember that amino acids are the building blocks of proteins.



- 4h. Describe the structure of a protein. (What elements do they contain?)
- 4i. Name 3 proteins that you eat every day.
- 4j. What is the difference in the chemical structure of monomers and polymers?

NUCLEOTIDES

4a. Remember that you should be using strategy

- 4b. _____ is the chemical that participates in a chemical reaction.
- 4c. _____ is the chemical that participates in a chemical reaction.
- 4d. _____ is the chemical that participates in a chemical reaction.
- 4e. _____ is the chemical that participates in a chemical reaction.
- 4f. _____ is the chemical that participates in a chemical reaction.
- 4g. _____ is the chemical that participates in a chemical reaction.
- 4h. _____ is the chemical that participates in a chemical reaction.
- 4i. _____ is the chemical that participates in a chemical reaction.
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- 4o. _____ is the chemical that participates in a chemical reaction.
- 4p. _____ is the chemical that participates in a chemical reaction.
- 4q. _____ is the chemical that participates in a chemical reaction.
- 4r. _____ is the chemical that participates in a chemical reaction.
- 4s. _____ is the chemical that participates in a chemical reaction.
- 4t. _____ is the chemical that participates in a chemical reaction.
- 4u. _____ is the chemical that participates in a chemical reaction.
- 4v. _____ is the chemical that participates in a chemical reaction.
- 4w. _____ is the chemical that participates in a chemical reaction.
- 4x. _____ is the chemical that participates in a chemical reaction.
- 4y. _____ is the chemical that participates in a chemical reaction.
- 4z. _____ is the chemical that participates in a chemical reaction.

Enzyme	Substrate
1. Amylase	Starch
2. Lipase	Triglyceride
3. Protease	Protein

Enzymes are proteins that act as catalysts. They speed up chemical reactions. Enzymes are made of amino acids. Amino acids are the building blocks of proteins. There are 20 different amino acids. Each amino acid has a central carbon atom bonded to a hydrogen atom, an amino group, a carboxyl group, and a variable group (R group). The variable group is what makes each amino acid unique.

- 4a. Describe the structure of an enzyme. (What elements do they contain?)
- 4b. Describe the structure of a substrate. (What elements do they contain?)
- 4c. Describe the structure of a catalyst. (What elements do they contain?)

4d. Enzymes are made of amino acids. Amino acids are the building blocks of proteins. There are 20 different amino acids. Each amino acid has a central carbon atom bonded to a hydrogen atom, an amino group, a carboxyl group, and a variable group (R group). The variable group is what makes each amino acid unique.



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Padhraic Smyth



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