

dengan $\gamma = \frac{C_p}{C_v}$ = konstanta Laplace, dan $\frac{C_p}{C_v} > 1$. C_p adalah kapasitas kalor gas pada tekanan tetap dan C_v adalah kalor gas pada volume tetap. Perhatikan diagram $p - V$ pada Gambar 9.7. Dari kurva hubungan $p - V$ tersebut, Anda dapat mengetahui bahwa:

- 1) Kurva proses adiabatik lebih curam daripada kurva proses isothermal.
- 2) Sama, tekanan, merupakan volume pada proses adiabatik tidak tetap.

Oleh karena sistem tidak melepaskan atau menerima kalor, pada kalor sistem proses adiabatik Q sama dengan nol. Dengan demikian, usaha yang dilakukan oleh sistem hanya mengulangi energi dalam sistem tersebut. Biasanya usaha pada proses adiabatik tersebut dinyatakan dengan persamaan berikut.

$$W = \frac{3}{2} nR(T_1 - T_2) = \frac{5}{2} (p_1 V_1 - p_2 V_2) \quad (9-8)$$

Contoh 9.6

Sebuah mesin memiliki rasio pemampatan $12 : 1$ yang berarti bahwa setelah pemampatan, volume gas menjadi $\frac{1}{12}$ volume awalnya. Anggap bahan bakar bercampur udara pada suhu 33°C , tekanan 1 atm, dan $\gamma = 1,4$. Jika proses pemampatan terjadi secara adiabatik, hitunglah tekanan pada keadaan akhir dan suhu campuran.

Jawab:

Diketahui: $V_2 = \frac{1}{12} V_1$, $T_1 = 33 + 273 = 306 \text{ K}$, dan $p_1 = 1 \text{ atm}$.

Untuk menentukan tekanan akhir p_2 , gunakan rumus:

$$p_1 V_1^\gamma = p_2 V_2^\gamma \rightarrow p_2 = p_1 \left(\frac{V_1}{V_2} \right)^\gamma \rightarrow p_2 = 1 \left(\frac{V_1}{\frac{1}{12} V_1} \right)^{1,4} = (12)^{1,4} = 32,4 \text{ atm.}$$

Suhu campuran akan akhir akhir T_2 diperoleh sebagai berikut:

$$T_1 V_1^{\gamma-1} = T_2 V_2^{\gamma-1} \rightarrow T_2 = T_1 \left(\frac{V_1}{V_2} \right)^{\gamma-1}$$

$$T_2 = 306 \text{ K} (12)^{1,4-1} = 306 \text{ K} (12)^{0,4} = 532 \text{ K} = 259^\circ\text{C}$$

Contoh 9.7

Udara sebesar $2 \times 10^3 \text{ J}$ diberikan secara adiabatik untuk memampatkan $0,5 \text{ mol}$ gas ideal monatomik, sehingga suhu mutlaknya menjadi 2 kali semula. Jika konstanta umum gas $R = 8,31 \text{ J/mol K}$, tentukanlah suhu awal gas.

Jawab:

Diketahui: $W = 2 \times 10^3 \text{ J}$, $T_2 = 2T_1$ dan $n = 0,5 \text{ mol}$.

$$W = \frac{5}{2} nR(T_2 - T_1) = \frac{5}{2} nR(2T_1 - T_1)$$

$$= \frac{5}{2} nR T_1$$

$$T_1 = \frac{2W}{5nR} = \frac{2(2 \times 10^3 \text{ joule})}{5(0,5 \text{ mol})(8,31 \text{ J/molK})} = 321 \text{ K}$$

Jadi, suhu awal gas adalah 321 K.



Gambar 9.7

Pada proses adiabatik, kurva $p - V$ lebih curam dibanding kurva dengan kurva $p - V$ pada proses isothermal.

Lompatan Kuantum



1 ... 2 atom thermal energy, absorbed when light (photons) having energy $h\nu$. When an electron (partikel) absorbs photons with energy $h\nu$, it jumps to a higher energy level (state) and emits light (photon) when it returns to a lower energy level.

Quantum Leap

The atom is in a lower energy state. When it absorbs a photon (light) having energy $h\nu$, it jumps to a higher energy state. When it returns to a lower energy state, it emits a photon (light) having energy $h\nu$.

Source: Wikipedia, April 2008

Conceptual Physics Fundamentals Answers

William S. Levine



Conceptual Physics Fundamentals Answers:

Fundamental Physics and Physics Education Research Burra G. Sidharth, Jesús Carnicer Murillo, Marisa Michelini, Carmen Perea, 2020-11-11 This book highlights selected contributions presented at the 15th annual international symposium Frontiers of Fundamental Physics FFP15 with the aim of informing readers about the most important recent advances in fundamental physics and physics education research The FFP series offers a platform for physicists from around the world to present their latest theories and findings The latest symposium was held in Orihuela Spain and covered diverse fields of research including gravitation astronomy and astrophysics physics of complex systems high energy physics and mathematical physics Considerable attention was also paid to physics education research teacher education in physics and the popularization of physics In a knowledge based society research into fundamental physics plays a vital role in both the advancement of human knowledge and the development of new technologies Presenting valuable new peer reviewed contributions submitted from 15 countries this book will appeal to a broad audience of scholars and researchers

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