

Note Taking Study Guide: Topic 1: Reconstruction

KEY

Goals	Actions / Events	Outcome
Return southern states to the Union.	Lincoln's Ten Percent Plan: a majority of voters in the rebelling states had to swear loyalty to the Union (this did not go into full effect as Lincoln was assassinated).	Southern states rejoin the Union.
Return southern states to the Union.	Johnson's Reconstruction Plan: a majority of voters in each southern state had to pledge loyalty to the Union and each state had to ratify the 13th amendment.	Southern states rejoin the Union.
Promote the rights of former slaves.	Congress and Lincoln created a combined Reconstruction Plan.	The Freedmen's Bureau: the federal government created and sponsored agency to promote the rights of former slaves by: -providing food and clothing to freedmen -assisted freedmen in finding employment -gave jobs to freedmen -provided medical care for one million people -created public schools for freedmen which laid the foundation for the South's public school system -also helped poor Southern whites
Promote the rights of	Congress and Johnson	The 13th amendment:

Note Taking Study Guide The Reconstruction Era

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Note Taking Study Guide The Reconstruction Era:

Whispering the Techniques of Language: An Psychological Quest through **Note Taking Study Guide The Reconstruction Era**

In a digitally-driven world wherever screens reign supreme and immediate communication drowns out the subtleties of language, the profound secrets and emotional subtleties hidden within phrases frequently go unheard. However, situated within the pages of **Note Taking Study Guide The Reconstruction Era** a fascinating literary prize blinking with natural feelings, lies a fantastic journey waiting to be undertaken. Published by an experienced wordsmith, that wonderful opus encourages viewers on an introspective trip, delicately unraveling the veiled truths and profound impact resonating within ab muscles material of each word. Within the emotional depths of this poignant evaluation, we shall embark upon a heartfelt exploration of the book is core subjects, dissect their fascinating publishing type, and yield to the powerful resonance it evokes strong within the recesses of readers hearts.

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apr 25 2017 this is understood when you consider the two chemicals appear unchanged before and after the reaction therefore the net ionic equation is $2\text{Cl}^- + 2\text{Ag}^+$

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ions that remain essentially unchanged during a reaction are called spectator ions this means you can ignore them when you write the ionic equation you only need to model how the solid

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the first step to writing a net ionic equation is balancing the chemical equation present let s use the reaction between sodium chloride and silver nitrate as an example next we write the

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highlights learning objectives by the end of this section you will be able to derive chemical equations from narrative descriptions of chemical reactions write and balance chemical

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the net ionic equation is the chemical equation that shows only those elements compounds and ions that are directly involved in the chemical reaction notice that in writing the net ionic

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this net ionic equation indicates that solid silver chloride may be produced from dissolved chloride and silver i ions regardless of the source of these ions these molecular and

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write the overall chemical equation the complete ionic equation and the net ionic equation for the reaction of aqueous silver fluoride with aqueous sodium phosphate to give solid silver

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the complete ionic equation for this reaction is as follows equation 4 13 2ag aq 2f aq 2nh₄ aq cr₂o₇²⁻ aq ag₂cr₂o₇ s 2nh₄ aq 2f aq because two nh₄ aq

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