

Summative Assessment: Nuclear Chemistry – Radioactivity & Nuclear Energy (Traditional Organization)

Matching: Match the description in column A with the correct term in column B. Write the letter in the blank provided. Each term matches with only one description, so be sure to choose the best description for each term.

Column A

- I 1. Used in the medical field to help provide information about how the body functions
- G 2. Helium nucleus
- C 3. Self-sustaining process
- M 4. Instrument used for measuring radioactivity levels
- D 5. Undergoes spontaneous nuclear decay
- J 6. High-energy photon
- A 7. Amount of radioactive material necessary for fission to occur
- H 8. Time it takes for half of a sample of radioactive nuclei to decay
- E 9. Splitting one heavy nucleus into two nuclei with smaller mass numbers
- H 10. Electron
- J 11. Particle with the same mass as an electron but a positive charge
- B 12. Subatomic particles found in the nucleus
- F 13. Combining two light nuclei to form a heavier nucleus
- K 14. The change of one element into another

Column B

- A. Critical Mass
- B. Nucleons
- C. Chain reaction
- D. Radioactive
- E. Fission
- F. Fusion
- G. Alpha particle
- H. Beta particle
- I. Gamma ray
- J. Positron
- K. Nuclear transmutation
- L. Radiotracers
- M. Geiger Counters
- N. Half-life

Multiple Choice - Choose the one best answer and write its letter in the left margin.

- C 15. Identify the missing particle: $^{239}_{94}\text{Pu} \rightarrow \underline{\hspace{1cm}} + ^4_2\text{He}$
- A. $^{235}_{92}\text{U}$
 B. $^{239}_{92}\text{Th}$
 C. $^{235}_{92}\text{U}$
 D. $^{239}_{96}\text{Am}$

- B 16. Identify the missing particle: $^{238}_{90}\text{Th} \rightarrow \underline{\hspace{1cm}} + ^4_2\text{He}$

Nuclear Chemistry Topic 12 Answer Key

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TABLE OF CONTENT

PART I Based on Class XI NCERT Some Basic Concepts of Chemistry Structure of Atom Classification of Elements and Periodicity in Properties Chemical Bonding and Molecular Structure States of Matter Thermodynamics Equilibrium Redox Reactions Hydrogen s Block Elements p Block Elements Organic Chemistry Some Basic Principles and Techniques Hydrocarbons Environmental Chemistry

PART II Based on Class XII NCERT The Solid State Solutions Electrochemistry Chemical Kinetics Surface Chemistry Nuclear Chemistry p Block Elements The d and f Block Elements Coordination Compounds Haloalkanes and Haloarenes Alcohols Phenols and Ethers Aldehydes Ketones and Carboxylic Acids Nitrogen Containing Compounds Biomolecules Polymers Chemistry in Everyday Life Analytical Chemistry General Principles and Processes of Isolation of Elements Questions Asked in JEE Main 2015 Solved Papers 2016 JEE Main BITSAT AP EAMCET TS EAMCET GGSIPU Solved Papers 2017 JEE Main Advanced BITSAT VIT WB JEE Solved Papers 2018 JEE Main Advanced BITSAT WB JEE Solved Papers 2019 JEE Main Advanced BITSAT WB JEE

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Table of Contents Nuclear Chemistry Topic 12 Answer Key

1. Understanding the eBook Nuclear Chemistry Topic 12 Answer Key
 - The Rise of Digital Reading Nuclear Chemistry Topic 12 Answer Key
 - Advantages of eBooks Over Traditional Books
2. Identifying Nuclear Chemistry Topic 12 Answer Key
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in a Nuclear Chemistry Topic 12 Answer Key
 - User-Friendly Interface
4. Exploring eBook Recommendations from Nuclear Chemistry Topic 12 Answer Key
 - Personalized Recommendations

- Nuclear Chemistry Topic 12 Answer Key User Reviews and Ratings
- Nuclear Chemistry Topic 12 Answer Key and Bestseller Lists
- 5. Accessing Nuclear Chemistry Topic 12 Answer Key Free and Paid eBooks
 - Nuclear Chemistry Topic 12 Answer Key Public Domain eBooks
 - Nuclear Chemistry Topic 12 Answer Key eBook Subscription Services
 - Nuclear Chemistry Topic 12 Answer Key Budget-Friendly Options
- 6. Navigating Nuclear Chemistry Topic 12 Answer Key eBook Formats
 - ePub, PDF, MOBI, and More
 - Nuclear Chemistry Topic 12 Answer Key Compatibility with Devices
 - Nuclear Chemistry Topic 12 Answer Key Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Nuclear Chemistry Topic 12 Answer Key
 - Highlighting and Note-Taking Nuclear Chemistry Topic 12 Answer Key
 - Interactive Elements Nuclear Chemistry Topic 12 Answer Key
- 8. Staying Engaged with Nuclear Chemistry Topic 12 Answer Key
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Nuclear Chemistry Topic 12 Answer Key
- 9. Balancing eBooks and Physical Books Nuclear Chemistry Topic 12 Answer Key
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Nuclear Chemistry Topic 12 Answer Key
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Nuclear Chemistry Topic 12 Answer Key
 - Setting Reading Goals Nuclear Chemistry Topic 12 Answer Key
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Nuclear Chemistry Topic 12 Answer Key
 - Fact-Checking eBook Content of Nuclear Chemistry Topic 12 Answer Key

- Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
- 14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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