



Middle Ages & Renaissance Study Guide

Middle Ages Dates: 500-1450 1. What was going historically during this era? What was life like? It was a chaotic period of social and political unrest. Religious and political differences between and within regions led to nearly constant warfare. The life of all the classes was dominated by the feudal system - feudalism. 2. Who or what had the power? The church 3. Who were learned or literate? Holy men or people of important status such as kings, queens, and knights most other people were illiterate 4. Who were allowed to sing in church? Only men were allowed to sing in church 5.

Most of the music that was notated during the middle ages was sacred or secular? Sacred 6. Was most medieval music vocal? Yes 7. What is Gregorian Chant? Discuss its origin, texture, melody, rhythm, text. How did it receive its name? What is its purpose? Music to which portions of the Roman Catholic service are sung by unaccompanied voices singing in unison. The melodies of Gregorian chant are commonly sung a capable and in unison by men and boys, or by women in female religious institutions such as convents. The rhythm is free and flexible.

The text may be treated in a syllabic manner with one note of music corresponding to each syllable of text. 8. When chant was notated, was the rhythm notated or just the melody? Gregorian Chant and just the melody 9. What are church modes? Any of a system of modes used in Gregorian chants up until 1600 10. What is the Mass? Roman Catholic worship service 11. What is the Proper of the Mass? Parts of the mass change according to the seasons of the church year around Easter and others at Characteristic; they are proper or appropriate, only at certain times. 12. What is the Ordinary of the Mass?

Parts of the mass celebrated at any season of the year or time of the day. 13. Who is Hildebrand of Bigger? 12th century Benedictine abbe who was a composer of sacred song and chant 14. When did composers begin to write polyphonic pieces? What was this early polyphony like? 15. What is a canon? Polyphonic composition in which all the voices perform the same melody, beginning at different times 16. What is a drone? Sustained tone 17. What is an station? Persistently repeated melodic or rhythmic pattern 18. What is the liturgy? Words of the mass 19. Who is Gallinule De Mach? The century poet and musician who composed the first complete polyphonic setting of the entire Ordinary of the Mass. 20. What city became the musical center of Europe during the mid to late Middle Ages? What was the musical center of this city and who worked there? Vienna Composers came from all over Europe to train in and around Vienna, and gradually they developed and formalized the standard musical forms that were to dominate European musical culture for the next several decades. 21. Composers began to write polyphonic songs that were not always based on chant, what were they based on instead? 22.

Middle Ages And Renaissance Study Guide Answers

**Alexander Fidora, Matthias Lutz-
Bachmann**



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web elements in the second row of the periodic table place their electrons in the $2n$ shell as well as the $1n$ shell for instance lithium li has three electrons two fill the $1s$ orbital and the third is placed in the $2s$ orbital giving an electron configuration of $1s^2 2s^1$

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web electron configurations and trends answer these discussion questions given the following part of an electron configuration $3s^2$ explain which part refers to the number of electrons the energy level and the sublevel what is the maximum number of electrons that can occupy a single orbital state and explain hund's rule

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web one more point needs to be emphasized about the relationship between electron configuration and the periodic table the atoms of elements in the same vertical column of the table have similar electron configurations for example consider the alkaline earth elements group iia

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web the electron removed during the ionization of beryllium $he\ 2s^2$ is an s electron whereas the electron removed during the ionization of boron $he\ 2s^2\ 2p^1$ is a p electron this results in a lower first ionization energy for boron even though its nuclear charge is greater by one proton

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web predictions from reliable sources have been used for these elements grayed out electron numbers indicate subshells filled to their maximum bracketed noble gas symbols on the left represent inner configurations that are the same in each period written out these are $he\ 2$ helium $1s^2$ $ne\ 10$ neon $1s^2\ 2s^2\ 2p^6$

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web q1 what is the electron configuration of the element potassium answer the atomic number of potassium is 19 a neutral potassium element has 19 electrons its electron configuration is $1s^2\ 2s^2\ 2p^6\ 3s^2\ 3p^6\ 4s^1$ or simply put $ar\ 4s^1$ q2 what is the total number of electrons that can be filled in s p and d subshells

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web by building up from hydrogen this table can be used to determine the electron configuration for any atom on the periodic table we will now construct the ground state electron configuration and orbital diagram for a selection of atoms in the first and second periods of the periodic table

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web electron configurations are a simple way of writing down the locations of all of the electrons in an atom as we know the positively charged protons in the nucleus of an atom tend to attract negatively charged electrons

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blocks depending on their electron configuration