

1. Which of the properties listed below is a characteristic of a liquid but not of a gas? B
A. It has the ability to flow.
B. It maintains its volume in an open container.
C. Its molecules are constantly in motion.
D. Its volume varies depending on temperature.
2. A 0.100 m³ chamber is filled with oxygen gas at standard conditions (101 kPa at 273 K). Assuming ideal conditions and behavior, what will be the pressure of the gas in the chamber after it is heated to 819 K? D
A. 34 kPa
B. 51 kPa
C. 202 kPa
D. 303 kPa
3. In 1898, J.J. Thomson provided the first hard evidence for the existence of subatomic particles in the atom by using a cathode ray tube to discover the C
A. proton.
B. neutron.
C. electron.
D. nucleus.
4. If all of the atoms in a substance have the same atomic number, the substance is a(an) B
A. compound.
B. element.
C. mixture.
D. alloy.
5. The balanced chemical equation for the formation of iron (III) oxide is B
A. $\text{Fe} + \text{O}_2 \rightarrow \text{FeO}_2$.
B. $4\text{Fe} + 3\text{O}_2 \rightarrow 2\text{Fe}_2\text{O}_3$.
C. $3\text{Fe} + \text{O}_2 \rightarrow \text{Fe}_3\text{O}_2$.
D. $\text{Fe} + \text{O} \rightarrow \text{FeO}$.

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